Effect of Water Activities of Heating and Recovery Media on Apparent Heat Resistance of Bacillus cereus Spores

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Spores of Bacillus cereus were heated and recovered in order to investigate the effect of water activity of media on the estimated heat resistance (i.e., the D value) of spores. The water activity (ranging from 0.9 to 1) of the heating medium was first successively controlled with three solutes (glycerol, glucose, and sucrose), while the water activity of the recovery medium was kept near 1. Reciprocally, the water activity of the heating medium was then kept at 1, while the water activity of the recovery medium was controlled from 0.9 to 1 with the same depressors. Lastly, in a third set of experiments, the heating medium and the recovery medium were adjusted to the same activity. As expected, added depressors caused an increase of the heat resistance of spores with a greater efficiency of sucrose with respect to glycerol and glucose. In contrast, when solutes were added to the recovery medium, under an optimal water activity close to 0.98, a decrease of water activity caused a decrease in the estimated D values. This effect was more pronounced when sucrose was used as a depressor instead of glycerol or glucose. When the heating and the recovery media were adjusted to the same water activity, a balancing effect was observed between the protective influence of the solutes during heat treatment and their negative effect during the recovery of injured cells, so that the overall effect of water activity was reduced, with an optimal value near 0.96. The difference between the efficiency of depressors was also less pronounced. It may then be concluded that the overall protective effect of a decrease in water activity is generally overestimated.

It has been recognized that the heat resistance of bacterial spores depends on the medium in which the spores are heated. The maximum thermostability of most microorganisms was found in the range of between 0.2 and 0.4 water activity (1, 3, 27, 28, 29). In typical ranges of water activities which are found in foodstuffs (a_{w} > 0.8), the heat resistance of microorganisms generally increases at decreasing water activities. However, the apparent effect of the water activity of the medium on spores or vegetative cells is complicated by the specific effect of solutes which are used as depressors. It is generally agreed that the occurrence of such solutes in the medium reduces the heat resistance of microorganisms. This antagonism between the protective effect of an increase in water activity and the opposite specific effect of depressors can explain conflicting data from various authors.

The influence of salt on the thermostability of microorganisms is disputed and depends on the heated type of microorganism. Some authors found no effect of the sodium chloride concentration on the heat resistance of bacteria (9, 29, 32, 42). Others observed a reduced heat resistance of microorganisms at increasing salt concentrations (7, 12, 22, 23). On the contrary, a protective effect of salt was found in several studies (6, 14, 26, 35, 38, 39, 40). Corry (14) deduced from his data that sodium chloride had a thermal protective effect on most heat-sensitive bacteria and the opposite effect on most heat-resistant species. Other solutes show the same opposite influence between their common depressor character which protects spores against heat and their specific effect which, on the contrary, reduces their heat resistance. It has been observed (21) that an increase of the thermal resistance of spores was more pronounced when the decrease of the medium water activity was generated by drying instead of an addition of glycerol, sodium chloride, lithium chloride, or glucose. Baird-Parker et al. (5) could not find any correlation between the heat resistance D (values) of salmonellae and the water activity of heating media when sodium chloride or glycerol were used as depressors. However, these researchers observed a clear protective effect of sucrose that was more pronounced for most heat-sensitive strains. It is generally recognized that sucrose is the most protective depressor, while glucose, sodium chloride, and lithium chloride show a clearly lower influence or even an opposite effect. Glycerol shows an intermediate behavior (13, 19, 20, 26, 37). Interactions between the influences of water activity and heating temperature were often observed. An increase of D values generated by a reduced water activity of the heating medium is generally related to an increase of z values. Moreover, several workers have demonstrated that the effect of the water activity of the heating medium depended on the treatment temperature; for example, in Staphylococcus epidermidis (39) or Listeria monocytogenes (37), the protective effect of decreasing water activity is more pronounced at a higher treatment temperature, while the opposite trend was observed for Staphylococcus aureus (38). A few predictive models describing the effect of the water activity of the heating medium on the heat resistance of spores were developed (8, 18, 31).

The nature of the recovery medium in which surviving heated cells are incubated has a great influence on their apparent heat resistance, i.e., their estimated D value (24). It is generally agreed that there is an optimum temperature of incubation for the cell ratio of recovery (16, 36) and the apparent D value (10). Acidification of the recovery medium...
TABLE 1. Predicted aw values of glycerol, glucose, and sucrose solutions at 25°C

<table>
<thead>
<tr>
<th>Molarity (M)</th>
<th>Glycerol aw</th>
<th>Glucose aw</th>
<th>Sucrose aw</th>
</tr>
</thead>
<tbody>
<tr>
<td>0.0</td>
<td>1.000</td>
<td>1.000</td>
<td>1.000</td>
</tr>
<tr>
<td>0.5</td>
<td>0.991</td>
<td>0.991</td>
<td>0.991</td>
</tr>
<tr>
<td>1.0</td>
<td>0.982</td>
<td>0.982</td>
<td>0.980</td>
</tr>
<tr>
<td>1.6</td>
<td>0.971</td>
<td>0.970</td>
<td>0.967</td>
</tr>
<tr>
<td>2.0</td>
<td>0.964</td>
<td>0.963</td>
<td>0.958</td>
</tr>
<tr>
<td>2.5</td>
<td>0.955</td>
<td>0.953</td>
<td>0.946</td>
</tr>
<tr>
<td>3.0</td>
<td>0.946</td>
<td>0.943</td>
<td>0.933</td>
</tr>
<tr>
<td>3.5</td>
<td>0.937</td>
<td>0.933</td>
<td>0.920</td>
</tr>
<tr>
<td>4.0</td>
<td>0.928</td>
<td>0.923</td>
<td>0.907</td>
</tr>
<tr>
<td>4.5</td>
<td>0.919</td>
<td>0.913</td>
<td>0.893</td>
</tr>
<tr>
<td>5.0</td>
<td>0.910</td>
<td>0.903</td>
<td>0.879</td>
</tr>
<tr>
<td>5.5</td>
<td>0.901</td>
<td>0.893</td>
<td>0.865</td>
</tr>
</tbody>
</table>

The strain *B. cereus* CNRZ 110 was obtained from the Institut National de Recherche Agronomique (Paris, France). Spores were kept in distilled water at 4°C. Cells were precultivated at 37°C for 5 days. Spores were then collected by scraping the surface of the agar, suspended in sterile distilled water, and washed three times by centrifugation (10,000 g for 15 min) (Bioblock Scientific model Sigma 3K30). The pellet was then resuspended in 5 ml of distilled water and 5 ml of ethanol. The obtained suspension was kept at 4°C for 12 h to eliminate vegetative nonsporulated bacteria and then washed again three times by centrifugation.

Lastly, the final suspension (ca. 10^10 spores ml^-1) was distributed into sterile Eppendorf microtubes and kept at 4°C.

**RESULTS**

**Effect of the water activity of the heating medium.** For identical heat treatment, the recovery conditions influence the apparent heat resistance of bacterial spores. A clear protective effect on spores of *B. cereus* heated at 95°C and at pH 7 was observed when solutes were added to the heating medium for the three types of depressors (Fig. 1 and 2). However, it can be seen that the effect of sucrose is more pronounced than that of glycerol and glucose. While a D95°C of about 4 min at a water activity close to 1 was found, at a water activity of 0.9 the observed D95°C became close to 11.2, 10.5, and 27 min for glycerol, glucose, and sucrose, respectively.

The three sets of data corresponding to each depressor were fitted according to the Gaillard et al. model (18) which, under isothermal conditions and at a fixed pH of the heating medium, can be reduced to:

\[
\log D_{(aw,1)} = \log D_{(1,1)} - \frac{aw - 1}{z_{aw}}
\]

where \( D_{(aw,1)} \) is the estimated D value at a water activity of the heating medium aw (which is the controlled variable) and a water activity of the recovery medium of 1. \( z_{aw} \) corresponds to the decrease in water activity of the treatment medium which would cause a 10-fold reduction of the decimal reduction time, with a water activity of the recovery medium of 1. The estimated parameter values are presented in Table 2.

**Effect of the water activity of the recovery medium.** Whatever the solute used as depressor in the recovery medium, heated spores show the same maximum apparent heat resistance (a D95°C value of ca. 5 min) at an optimum water activity close to 0.98 (see Fig. 3 and 4). Under this optimal value, an increasing concentration of the three depressors causes a de-
crease in the apparent heat resistance of the spores. Sucrose presents the most pronounced effect, followed in turn by glucose and glycerol. At water activity of 0.92, the estimated $D_{95°C}$ values were 0.9, 1.9, and 2.5 min with sucrose, glucose, and glycerol, respectively.

We tried to adapt the model describing the influence of the pH of the recovery medium to the apparent thermal resistance of spores, which was tested in our laboratory (15) by substituting pH for water activity, leading to the following equation:

$$\log D_{(1,aw)} = \log D_{(1,1)} - \left(\frac{a'_{aw} - a'_{aw opt}}{z_{aw}^*}\right)^2$$

where $D_{(1,aw)}$ is the estimated $D$ value at a water activity of the recovery medium $a'_{aw}$ (which is the controlled variable) and a water activity of the heating medium of 1. $z_{aw}^*$ corresponds to the decrease of water activity of the recovery medium which would cause a 10-fold reduction of the decimal reduction time, with a water activity of the heating medium of 1. The estimated parameter values are presented in Table 3.

Overall effect of water activity of foods on the apparent heat resistance of spores. Since foods make up both the heating medium and the recovery medium, a third set of experiments was carried out in which spores were recovered at the same water activity as those of the heating menstruum (Fig. 5). With respect to the second set of data in which the water activity of the heating medium was kept to 1, some noteworthy differences appear. First, the overall influence of water activity becomes relatively slight, while the differences in the curve patterns according to the depressors used are less pronounced than those of Fig. 4. Second, a shift of the optimum water activity from 0.98 toward 0.96 can be observed, with a maximum $D_{95°C}$ value of close to 8 min instead of 5 min. Equations 1 and 2 cannot directly be combined in order to build a model.
which would take into account the overall effect of the food water activity because they were developed by keeping the water activity of the recovery medium at 1 for equation 1, and keeping the water activity of the heating medium at 1 for equation 2. The accuracy of this third set of data was too poor to allow suitable modeling.

**DISCUSSION**

It has been confirmed that an increase in the water activity produces opposite effects on the apparent heat resistance of spores according to whether it concerns the heat treatment or the recovery medium. Inside the investigated water activity range (0.9 to 1), which corresponds to that of most typical foods, a clear protective effect of an increasing water activity of the heating medium can be observed. For a fixed water activity, the degree of protection depends on the type of depressor used; according to our investigations, sucrose showed a more effective protective effect than glycerol and glucose, a finding which is in agreement with observations by other authors (5, 19, 26, 37). As in the case of NaCl, the protective influence of an increase of the water activity could partly be balanced by a specific antagonistic and toxic effect of glycerol and glucose. Moreover, the plasmolyse which is partly responsible for the heat protection of spores is limited by the penetration of glycerol and glucose inside the cells. This limitation does not exist when the depressor is not uptaken inside cells, which is the case with sucrose. Another explanation for the protective effect of a depressor in the heating medium could be an inhibition of spore germination. Anagnastopoulos and Sidhu (4) observed for *Bacillus stearothermophilus* that the percentage of germination decreases when the water activity decreased and that, at the same water activity, the percentage of germination was lower in a nutrient broth supplemented with sucrose than in a broth supplemented with glycerol. Hydration of spore protoplasm is an important condition of spore activation and initiation of germination. Germination is inhibited in the absence of moisture or in a concentrated solution of nonpenetrating solute. The spores which do not germinate are protected during heat treatment and can germinate and grow during recovery. This explanation is in agreement with our results: when the water activity decreases, the heat resistance of spores increases and, at the same water activity, sucrose, a less-penetrating solute in protoplast than glycerol and glucose, shows a greater protective effect than these two solutes. Moreover, according to Anagnastopoulos and Sidhu, the \( z_{aw} \) values for glycerol and sucrose (0.28 and 0.13, respectively) correspond to the \( z_{aw} \) value difference which leads to a 10-fold reduction of the percentage of germination of *B. stearothermophilus* at 75°C with glycerol and sucrose (0.31 and 0.11, respectively).

It is recognized that the addition of sodium chloride to the recovery medium causes both a reduction of viability of cells

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**TABLE 3. Parameter estimates of equation 2**

<table>
<thead>
<tr>
<th>Depressor</th>
<th>( D_{11} ) (min)</th>
<th>( a_{aw} )</th>
<th>( z_{aw} ) (range)</th>
<th>( R^2 )</th>
</tr>
</thead>
<tbody>
<tr>
<td>Glycerol</td>
<td>4.75</td>
<td>0.978</td>
<td>0.103 (0.098–0.108)</td>
<td>0.893</td>
</tr>
<tr>
<td>Glucose</td>
<td>4.55</td>
<td>0.990</td>
<td>0.113 (0.106–0.121)</td>
<td>0.883</td>
</tr>
<tr>
<td>Sucrose</td>
<td>4.52</td>
<td>0.978</td>
<td>0.068 (0.066–0.070)</td>
<td>0.941</td>
</tr>
</tbody>
</table>

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**FIG. 4.** (A) *B. cereus* \( D_{aw\,5°C} \) value versus the \( a_w \) of the recovery medium adjusted with glycerol. Symbols: ●, experimental data; —, calculated values according to equation 2. (B) *B. cereus* \( D_{aw\,5°C} \) value versus the \( a_w \) of the recovery medium adjusted with glucose. Symbols: ■, experimental data; —, calculated values according to equation 2. (C) *B. cereus* \( D_{aw\,5°C} \) value versus the \( a_w \) of the recovery medium adjusted with sucrose. Symbols: ▲, experimental data; —, calculated values according to equation 2.
and a lower apparent D value of the spores (7, 12, 16, 22, 32). However, as far as we know, the influence of the water activity of the recovery medium and the types of depressors used upon the estimated D values of spores had never been investigated. In our experimental conditions, a maximal apparent heat resistance of spores appeared at an optimal water activity close to 0.98. Below this value, a decrease in the water activity of the recovery medium causes a reduction of the estimated D value of the spores. It is interesting to note regarding this trend that the depressors appear at the same increasing order of effectiveness as was found regarding their protective effect in the heating medium: glycerol, glucose, and sucrose, respectively. This observation is consistent with the behavior of germinated (i.e., non-refractile) spores, which are rapidly permeable to glycerol, permeable to glucose by active transport, and permeable to sucrose to a lower extent. This order also corresponds to the increasing order of molecular weights and to the decreasing degree of penetration inside the cells. Particularly, the absence of uptake of sucrose by cells keeps a sharp gradient of osmotic pressure between the cell inside and the outside medium, which reduces the viability of surviving cells.

The overall influence of water activity of a single medium which makes up both the heating menstruum and the recovery medium upon the apparent D values of spores had not yet been investigated. Our results show that the protective effect of a decrease in the water activity of the medium during heat treatment is more or less offset by a reduction of the viability of surviving cells during the recovery. Moreover, because depressors, which are most effective in the heat protection of spores, are also responsible for the maximum loss of viability of injured cells, their overall difference in efficiency is greatly reduced. Whereas most authors have determined the optimal water activity of the maximum heat resistance of spores to be between 0.2 and 0.4 because the surviving cells were recovered under optimal conditions when the media of the heat treatment and of the incubation are the same, as for heat-processed foods, the optimal water activity is actually near 0.96.

Investigating separately the influence of water activity of the heating medium on the heat resistance of spores on the one hand and the effect of water activity of the recovery medium on the viability of surviving cells on the other hand obviously provides very interesting and useful data: these two effects must be regarded as two different factors that interact with each other. However, when the heating medium is also the recovery medium, it is worth investigating the overall influence of water activity on the apparent heat resistance of the spores, which reflects both their immediate thermal resistance during heating and their ability to grow in a recovery medium.

In the framework of predictive microbiology, we could develop two separate models for describing the effect of the water activity of the heating menstruum on the one hand and that of water activity of the recovery medium on the other hand upon the apparent heat resistance of the spores. However, further works would be needed in order to develop an overall predictive model adapted for heat-processed food calculations. Such a model would take into account both opposite effects of water activity and would allow us to improve of heat treatment optimization.
REFERENCES