Isolation of a Novel Cutinase Homolog with Polyethylene Terephthalate-Degrading Activity from Leaf-Branch Compost by Using a Metagenomic Approach

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The gene encoding a cutinase homolog, LC-cutinase, was cloned from a fosmid library of a leaf-branch compost metagenome by functional screening using tributyrin agar plates. LC-cutinase shows the highest amino acid sequence identity of 59.7% to Thermomonospora curvata lipase. It also shows the 57.4% identity to Thermobifida fusca cutinase. When LC-cutinase without a putative signal peptide was secreted to the periplasm of Escherichia coli cells with the assistance of the pelB leader sequence, more than 50% of the recombinant protein, termed LC-cutinase*, was excreted into the extracellular medium. It was purified and characterized. LC-cutinase* hydrolyzed various fatty acid monoesters with acyl chain lengths of 2 to 18, with a preference for short-chain substrates (C4 substrate at most) most optimally at pH 8.5 and 50°C, but could not hydrolyze olive oil. It lost activity with half-lives of 40 min at 70°C and 7 min at 80°C. LC-cutinase* had an ability to degrade poly(ε-caprolactone) and polyethylene terephthalate (PET). The specific PET-degrading activity of LC-cutinase* was determined to be 12 mg/h/mg of enzyme (2.7 mg/h/mkat of pNP-butyrate-degrading activity) at pH 8.0 and 50°C. This activity is higher than those of the bacterial and fungal cutinases reported thus far, suggesting that LC-cutinase* not only serves as a good model for understanding the molecular mechanism of PET-degrading enzyme but also is potentially applicable for surface modification and degradation of PET.

Cutinase (EC 3.1.1.74) is a lipolytic/esterolytic enzyme that hydrolyzes not only cutin, which is a major component of plant cuticle (38), but also water-soluble esters and insoluble triglycerides (12). It hydrolyzes these substrates to carboxylic acids and alcohols through the formation of an acyl enzyme intermediate, in which the active-site serine residue is acylated by the substrate. This serine residue is located within a GXSXG sequence motif and forms a catalytic triad with His and Asp. Cutinase has been found in both fungi and bacteria. The crystal structures of two fungal cutinases from Fusarium solani f. sp. pisi (22) and Glomerella cingulata (27) have been determined. According to these structures, cutinase shares a common α/β hydrolase fold with lipase and esterase (28). However, cutinase, like esterase, does not have a lid structure, which is responsible for interfacial activation of lipase (8). Therefore, cutinase does not show interfacial activation like esterase (14). Cutinase has recently received much attention because of its potential application for surface modification and degradation of aliphatic and aromatic polysteres (16), especially polyethylene terephthalate (PET), which is a synthetic aromatic polyester composed of terephthalic acid (TPA) and ethylene glycol (10, 16, 36, 39). However, the number of cutinases, which have been studied regarding PET modification, is still limited, and this limitation may result in the delay of the research toward the practical use of cutinases. Therefore, isolation of a novel cutinase with PET-degrading activity is needed.

Metagenomics is the study of genetic material recovered directly from environmental sources (17, 30). Because more than 99% of microorganisms in nature cannot be cultivated by the conventional method (3), metagenomics has attracted many researchers, who intend not only to increase our knowledge on protein sequence space in nature but also to isolate novel enzymes with potentially useful application. By using this approach, a variety of novel enzymes, including lipases/esterases, cellulases, and proteases, have been isolated and characterized (33–35).

Microorganisms that can degrade plant cell wall produce a variety of plant cell wall-degrading enzymes, which include not only carbohydrate-degrading enzymes but also lipolytic/esterolytic enzymes. For example, the plant pathogenic bacterium Xanthomonas oryzae secretes an esterase, LipA, which is involved in degradation of cell walls in a synergetic manner with other cell wall-degrading enzymes (5). In EXPO Park, Japan, leaves and branches cut from the trees are collected periodically, mixed with urea, and agitated for composting. The temperature increases up to ~70°C inside this compost (leaf-branch compost) and then decreases to ~50°C roughly 1 year later upon completion of composting. This compost is expected to be rich in various plant cell wall-degrading microorganisms and therefore is a promising source of the genes encoding novel enzymes with cutinase activity.

In the present study, we constructed a DNA library for metagenomic study from leaf-branch compost and performed function-based screening for the genes encoding lipolytic/esterolytic enzymes using an agar medium containing tributyrin. We
identified the gene encoding a novel cutinase homolog, termed LC-cutinase, which shows an amino acid sequence identity of 57.4% to cutinase from *Thermobifida fusca*, overexpressed it in *Escherichia coli*, and purified and characterized the recombinant protein. We show that LC-cutinase exhibits a PET-degrading activity and is a potent candidate for applications in various industrial fields, especially in the textile industries.

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**MATERIALS AND METHODS**

Cells, plasmids, and enzymes. *E. coli* BL21-CodonPlus(DE3)-RP ([F-ompT lacIq (rpsL*)Δ (λ gal (DE3) endA Hsd (argU prol Cam*)]) was obtained from Stratagene (La Jolla, CA). Plasmid pET25b was purchased from Novagen (Madison, WI). *E. coli* BL21-CodonPlus(DE3)-RP transformants were grown in lysogeny broth (LB) medium (10 g of tryptone, 5 g of yeast extract, and 10 g of NaCl in 1 liter of H2O) supplemented with 50 mg of ampicillin liter−1. *Burkholderia cepacia* Bc-Lip, and 0.013 and 0.50 Cr-Lip with 50 mg of ampicillin liter−1. *Candida rugosa* Bc-Lip with 50 mg of ampicillin liter−1, and 0.013 and 0.50 Cr-Lip, 50 mg of ampicillin liter−1, and 0.013 and 0.50 Cr-Lip, with 50 mg of ampicillin liter−1.

**Construction of DNA library and screening.** The compost sample was taken from the core (1 m below the surface) of the 4-month-old leaf-branch compost made from leaves and branches in EXPO Park, Japan. The temperature and pH of this leaf-branch compost are 67°C and pH 7.5. DNA from metagenomic study was extracted from this compost sample using a CopyControl fosmid library (Sapporo, Hokkaido, Japan). The protein was dissolved in 10 mM Tris-HCl (pH 8.0) and cultivation was continued overnight. The LC-cutinase[36–293] derivative, termed LC-cutinase*, was purified from the culture supernatant at 4°C as described below. The LC-cutinase[36–293] derivative, termed LC-cutinase*, was supplemented with 50 mg of ampicillin liter−1.

The production level of the protein in *E. coli* cells and the purity of the protein were analyzed by sodium dodecyl sulfate-polyacrylamide gel electrophoresis (SDS-PAGE) (20) using a 12% polyacrylamide gel, followed by staining with Coomassie brilliant blue (CBB). The amount of the protein was estimated from the intensity of the band visualized by CBB staining using the Scion Image program. The N-terminal amino acid sequence of the protein was determined by a Procise automated sequencer model 491 (Applied Biosystems). The protein concentration was determined from the UV absorption on the basis that the absorbance of a 0.1% (1.0 mg/ml) solution at 280 nm is 1.37. This value was calculated by using $e = 5,252 \text{ M}^{-1} \text{cm}^{-1}$ for tyrosine and 5,252 $\text{ M}^{-1} \text{cm}^{-1}$ for tryptophan at 280 nm.

**Construction of mutant protein.** The pET25b derivative for overproduction of S165A-cutinase*, in which the active-site serine residue, Ser165, of LC-cutinase* is replaced by Ala, was constructed by PCR using the QuikChange II site-directed mutagenesis kit (Stratagene). The mutagenic primers were designed such that the TCG codon for Ser165 is changed to GCG for Ala. The mutant protein was overproduced and purified as described above for LC-cutinase*.

**Circular dichroism spectra.** The far-UV circular dichroism (CD) spectra were measured at 25°C on a J-725 spectropolarimeter (Japan Spectroscopic, Tokyo, Japan). The protein was dissolved in 10 mM Tris-HCl (pH 7.0) containing 1 mM DTT. The protein concentration was 0.1 mg/ml and a cell with an optical path length of 2 mm was used. The mean residue ellipticity (θ, degrees cm² dmol⁻¹) was calculated using an average amino acid molecular mass of 110.

**Enzyme assays.** The enzyme activity was determined at the temperature indicated in 1 ml of 25 mM Tris-HCl (pH 8.0) containing 2% acetonitrile and 1 mM p-nitrophosphoryl butyrate ($C_4$). The amount of p-nitrophenol (pNP) released from the substrate was determined from the absorption at 405 nm with an absorption coefficient of 18,600 $\text{M}^{-1} \text{cm}^{-1}$ by automatic spectrophotometer (Hitachi spectrophotometer U-2810; Hitachi High-Technologies, Tokyo, Japan). One katal unit of enzymatic activity was defined as the amount of enzyme that produced 1 mol of pNP per s. The specific activity was defined as the enzymatic activity per milligram of protein.

For determination of substrate specificity, pNP monoesters of fatty acids with acyl chain lengths of 2 (pNP-acetate), 4 (pNP-butyrate), 6 (pNP-hexanoate), 8 (pNP-caprylate), 12 (pNP-laureate), 14 (pNP-myristate), 16 (pNP-palmitate), and 18 (pNP-stearate) (Sigma) and olive oil were used as a substrate. Measurement of the enzyme activities for hydrolyses of pNP monoesters of fatty acids was performed as described above for hydrolysis of pNP-butyrate, except that the reaction mixture contained 25 mM Tris-HCl (pH 8.0), 10% acetonitrile, and 1 mM substrate. One katal unit of enzymatic activity was defined as the amount of enzyme that produced 1 mol of pNP per s. Measurement of the enzymatic activity for hydrolysis of olive oil was performed at the temperature indicated by titrating the liberated fatty acid with 10 mM NaOH as described previously (2), except that the reaction was carried out in 25 mM Tris-HCl.
TABLE 1 Proteins with high amino acid sequence identities to LC-cutinase as determined by BLAST searching

<table>
<thead>
<tr>
<th>Enzyme</th>
<th>No. of residues</th>
<th>Source organism</th>
<th>Accession no.</th>
<th>Identity (%)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Triacylglycerol lipase</td>
<td>289</td>
<td>Thermomonospora curvata</td>
<td>YP_003298899</td>
<td>59.7</td>
</tr>
<tr>
<td>Putative lipase</td>
<td>334</td>
<td>Streptomyces ambofaciens</td>
<td>CAJ88461</td>
<td>57.8</td>
</tr>
<tr>
<td>Cutinase</td>
<td>261</td>
<td>Thermobifida fusca</td>
<td>YP_288944</td>
<td>57.4</td>
</tr>
<tr>
<td>Lipase</td>
<td>310</td>
<td>Streptomyces coelicolor A3(2)</td>
<td>AAD09315.1</td>
<td>57.4</td>
</tr>
<tr>
<td>Lipase</td>
<td>304</td>
<td>Saccharomonospora viridis</td>
<td>YP_00314604</td>
<td>55.8</td>
</tr>
<tr>
<td>Lipase</td>
<td>304</td>
<td>Steptomyces albus</td>
<td>LPZ_04702335.1</td>
<td>55.4</td>
</tr>
<tr>
<td>Cutinase</td>
<td>290</td>
<td>Nocardiosis dassonvillii</td>
<td>ZP_0433106.1</td>
<td>54.7</td>
</tr>
<tr>
<td>Lipase</td>
<td>262</td>
<td>Streptomyces exfoliatus</td>
<td>1JFR_A</td>
<td>53.9</td>
</tr>
</tbody>
</table>

* The amino acid sequence identities of proteins to LC-cutinase without putative signal peptide.

(pH 8.0). One katal unit of enzymatic activity was defined as the amount of enzyme that liberated 1 mol of fatty acid per s.

Inhibition with E600. LC-cutinase* (1.0 nmol) was incubated at 50°C for 30 min in 100 μl of 10 mM Tris-HCl (pH 7.6) containing 5 mM diethyl pNP phosphate (E600; Sigma). The residual activity was determined at 30°C using pNP-butyrate as a substrate.

Detection of PCL-degrading activity. Poly(e-caprolactone) (PCL)-degrading activity was determined by measuring the weight loss of a PCL film after incubation with the enzyme. For preparation of PCL film, 20 to 30 mg of PCL (Wako Pure Chemical, Osaka, Japan) was melted at 80°C and pressed into a thin film with ~5 mm in diameter at room temperature. This PCL film was added into 1 ml of 500 mM Tris-HCl (pH 8.0) and preincubated at 50°C for 5 min. The reaction was initiated by the addition of enzyme (5 μg for LC-cutinase* and 50 μg for Bc-Lip and Cr-Lip) and continued at 50°C for 6 h. After incubation, the films were washed with water and ethanol and dried for measurement of the weight loss.

Detection of PET-degrading activity. PET-degrading activity was determined by measuring the weight loss of a PET film after incubation with the enzyme. For preparation of PET film, a plastic package made of PET was cut into ~5-mm² pieces (20 to 25 mg per piece). This PET film was added into 1 ml of 500 mM Tris-HCl (pH 8.0) and preincubated at 50°C for 5 min. The reaction was initiated by the addition of enzyme (5 μg for LC-cutinase* and 50 μg for Bc-Lip and Cr-Lip) and continued at 50°C with gentle shaking for 24 h. After incubation, the films were washed with water and ethanol and dried for measurement of the weight loss. PET-degrading activity was also determined by quantifying the fatty acids released upon hydrolysis of PET with 50 mM NaOH. A PET film was incubated with the enzyme as mentioned above, except that the concentration of the reaction buffer was reduced to 100 mM and the incubation time was changed to 12 and 30 h.

Detection of cutin-degrading activity. Cutin fibers were prepared from tomato peels as described previously (23). These fibers (10 mg) were added into 1 ml of 20 mM Tris-HCl (pH 8.0) and preincubated at 50°C for 5 min. The reaction was initiated by the addition of enzyme (50 μg for LC-cutinase* and 100 μg for Bc-Lip and Cr-Lip) and continued at 50°C with gentle shaking. At appropriate intervals, an aliquot of the reaction mixture was withdrawn, and the fatty acids released upon hydrolysis of cutin were quantified by titration with 20 mM NaOH.

Nucleotide sequence accession number. The nucleotide sequence of the LC-cutinase gene has been deposited in GenBank under accession number HQ704839.

RESULTS AND DISCUSSION

Gene cloning of lipolytic/esterolytic enzymes from DNA library for metagenomic study. Extraction of DNA from 4-month-old leaf-branch compost (2.5 g) yielded 11 μg of high-molecular-weight DNA for metagenomic study. Ligation of sheared DNA into fosmid vector (pCC1FOS), followed by transformation of E. coli cells (E. coli EPI300-T1R), produced the DNA library for metagenomic study with a library size of approximately 2.1 × 10⁴ CFU. The restriction fragment length polymorphisms of 10 randomly selected clones using BamHI restriction enzymes showed nonredundant patterns and an average insert size of 35 kb.

Screening of the library for genes encoding lipolytic/esterolytic enzymes was performed using tributyrin agar plates. E. coli transformants which gave a halo should contain these genes. Of approximately 6,000 clones screened, 19 clones gave a halo on tributyrin agar plates when they were incubated at 37°C for 3 days. Three of them, which gave the largest halo at 50°C, were chosen to determine the nucleotide sequences of the genes responsible for halo formation. Determination of the nucleotide sequences of these genes by transposon mutagenesis indicates that all three clones harbor the same gene encoding a lipolytic/esterolytic enzyme. This protein is termed LC-cutinase (cutinase homolog from leaf-branch compost). LC-cutinase is composed of 293 amino acid residues with a calculated molecular mass of 31,494 and an isoelectric point (pI) of 9.3.

Amino acid sequence. Similarity search using blastp program indicates that LC-cutinase shows relatively high amino acid sequence identities of 54 to 60% to lipases, which have been classified as family III lipases (7), and cutinases (Table 1). It shows the highest amino acid sequence identity of 59.7% to Thermomonospora curvata lipase. The amino acid sequence of LC-cutinase is compared to those of Thermobifida fusca cutinase and Streptomyces exfoliatus lipase in Fig. S1 in the supplemental material. T. fusca cutinase shows the highest amino acid sequence identity of 57.4% to LC-cutinase among various cutinases and has been biochemically characterized as a representative member of bacterial cutinases (9). S. exfoliatus lipase is the only protein listed in Table 1, for which the crystal structure is available (37).

Analysis of the amino acid sequence of LC-cutinase using SMART (http://smart.embl.de) (21, 31) suggests that LC-cutinase is a secretory protein and has a 34-residue signal peptide at the N terminus. The mature region of LC-cutinase without this putative signal peptide, LC-cutinase[35-293], is composed of 259 amino acid residues with a calculated molecular mass of 27,902 and an isoelectric point (pI) of 9.3. This size is similar to those of T. fusca cutinase and S. exfoliatus lipase, both of which are composed of 261 amino acid residues. Three active-site residues (Ser165, Asp210, and His242) that form a catalytic triad and two residues ( Tyr95 and Met166) whose main-chain amide groups form an oxyanion hole are fully conserved in the LC-cutinase sequence (see Fig. S1 in the supplemental material). A typical pentapeptide GxSxG sequence motif is also fully conserved in the LC-cutinase sequence.

Overproduction of LC-cutinase. Met-Asp-LC-cutinase[36-293] was overproduced in E. coli as a fusion protein with the pelB leader sequence. Upon overproduction, the recombinant protein not only accumulated in the cells in a soluble form but also was released in...
the extracellular medium (data not shown). With the release of this recombinant protein, a variety of the proteins, presumably periplasmic proteins, were released in the extracellular medium (data not shown), suggesting that the recombinant protein is released in the extracellular medium due to a leakage. The production level of the recombinant protein accumulated in the cells and that released in the extracellular medium are estimated to be 6 and 8 mg/liter of culture, respectively, from the intensities of the bands visualized by CBB staining following SDS-PAGE. The molecular masses of these proteins estimated from SDS-PAGE are identical to each other (29 kDa). This value is slightly higher than but comparable to the calculated one of Met-Asp-LC-cutinase[36-293] (28,063 Da). These results suggest that Met-Asp-LC-cutinase[36-293] is translocated across the cytoplasmic membrane by the Sec transport system with the assistance of the pelB leader sequence, accumulated in the periplasmic space, and more than 50% of it was released in the extracellular medium due to a leakage of the outer membrane. This leakage mechanism of LC-cutinase* remains to be elucidated.

It should be noted that LC-cutinase with its own signal peptide accumulated in E. coli cells in inclusion bodies without cleavage of signal peptide upon overproduction (data not shown). This result suggests that this signal peptide cannot work properly in E. coli cells.

**Purification of LC-cutinase**. Because of the ease of the purification procedure, the recombinant protein was purified from the extracellular medium to give a single band on SDS-PAGE (see Fig. S2 in the supplemental material). The amount of the protein purified from 1-liter culture was roughly 6 mg. The N-terminal amino acid sequence of the purified protein was determined to be Gln-Pro-Ala-Met, indicating that the C-terminal five residues of the pelB leader sequence and the subsequent Met-Asp sequence are kept attaching to the N terminus of LC-cutinase[36-293]. This result indicates that the 22-residue pelB leader sequence is cleaved by signal peptidase at the peptide bond between the fifth and sixth residues from the C terminus, when Met-Asp-LC-cutinase[36-293] is translocated across the cytoplasmic membrane. This incompletely processing of signal peptide has also been reported for T. fusca cutinase (11). Upon overproduction of the OmpA-cutinase fusion protein in E. coli cells, this protein is transported to the periplasm after cleavage of the OmpA sequence. However, the OmpA sequence is not uniformly cleaved by signal peptidase. It is cleaved at multiple sites, mainly at the peptide bond between the ninth and tenth residues from the C terminus. It has been suggested that overload of Sec-transportation pathway forces cleavage of signal peptide at incorrect sites (11). The LC-cutinase[36-293] derivative with an N-terminal seven-residue extension (Gln-Pro-Ala-Met-Ala-Met-Asp) is termed LC-cutinase*. LC-cutinase* is composed of 265 amino acid residues with a calculated molecular mass of 28,517 Da.

**Enzymatic activity of LC-cutinase**. The pH dependence of LC-cutinase* was analyzed at various pH ranging from 5.5 to 9.5 and 30°C using pNP-butyratc as a substrate. LC-cutinase* exhibited the highest activity at pH 8.5 (specific activity of 3.3 ± 0.45 μkat/mg) and roughly 70% of the maximal activity at pH 7.0 (sodium phosphate) and pH 9.5 (see Fig. S3A in the supplemental material). The temperature dependence of LC-cutinase* was analyzed at various temperatures ranging from 30 to 80°C and pH 7.0 (sodium phosphate) using pNP-butyratc as a substrate. LC-cutinase* exhibited the highest activity at 50°C (specific activity of 3.2 ± 0.35 μkat/mg) and roughly 70% of the maximal activity at 30 and 70°C (Fig. S3B in the supplemental material). These results indicate that the optimum pH and temperature for enzymatic activity of LC-cutinase* are pH 8.5 and 50°C. The specific activity of LC-cutinase* at the optimum condition was 4.5 ± 0.52 μkat/mg. This activity was not changed either in the presence of 10 mM CaCl₂ or 10 mM EDTA, suggesting that LC-cutinase* does not require divalent metal ions for activity. It has been reported that the specific activity of T. fusca cutinase is 6.0 μkat/mg for tributyrin (C₃) (19) and 7.6 μkat/mg for pNP-butyratc (C₄) at pH 8.0 and 60°C (9). Thus, the specific activity of LC-cutinase* for pNP-butyratc is slightly lower than but comparable to that of T. fusca cutinase.

Substrate specificity of LC-cutinase* was analyzed using olive oil and various pNP monoesters of fatty acids with acyl chain lengths of 2 to 18 as a substrate at pH 8.0 and 37°C. The enzymatic activity was determined at this condition, instead of the optimum condition, because the stability of certain substrates, such as pNP-acetate, decreases as the pH and temperature increase beyond pH 8.0 and 40°C. The specific activity determined at pH 8.0 and 37°C was roughly 80% of that determined at the optimum condition. The specific activities of LC-cutinase* for various pNP-substrates relative to that for pNP-butyratc are shown in Fig. 1. LC-cutinase* hydrolyzed pNP-butyratc (C₄) most preferentially among various pNP substrates examined. It hydrolyzed pNP-hexanoate (C₆) and pNP-caprylate (C₁₄) with comparable efficiencies. However, it hydrolyzed substrates with acyl chain lengths of ≥12 with much lower efficiencies. Its activities for these substrates decrease as the acyl chain lengths of these substrates increase. It did not hydrolyze olive oil at 30°C. Thus, LC-cutinase* shows strong preference to the substrates with short acyl chain length. Preference to short-chain substrates has also been reported for T. fusca cutinase (19).

To examine whether LC-cutinase* is inhibited by diethyl pNP-phosphate (E600), it was incubated at pH 7.6 and 50°C for 30 min in the presence of E600. Upon incubation with E600, LC-cutinase* completely lost the enzymatic activity for pNP-butyratc, indicating that it is inhibited by E600.

**Stability.** To analyze the stability of LC-cutinase* against irreversible heat inactivation, LC-cutinase* (0.1 mg/ml) was incubated in 25 mM Tris-HCl (pH 8.0) containing 10% acetonitrile using pNP-acetate (C₃), pNP-butyratc (C₄), pNP-hexanoate (C₆), pNP-caprylate (C₁₄), pNP-laurate (C₁₂), pNP-myristate (C₁₄), pNP-palmitate (C₁₄), or pNP-stearate (C₁₈) as a substrate. The specific activities relative to that determined for hydrolysis of pNP-butyratc are shown. The experiment was carried out at least twice, and the average values are shown together with error bars.
bated in 10 mM sodium phosphate (pH 7.0) at 50, 60, 70, and 80°C. With appropriate intervals, an aliquot of the solution was withdrawn and analyzed for residual activity at 30°C using pNP-butyrate. As shown in Fig. S4 in the supplemental material, LC-cutinase* lost activity with half-lives of 5 h at 50°C, 80 min at 60°C, 40 min at 70°C, and 7 min at 80°C. It has been reported that T. fusca cutinase exhibits the highest activity at 60°C and pH 8, and high stability with residual activity of over 50% after 40 h at 60°C (9). Thus, LC-cutinase* is slightly less stable than T. fusca cutinase.

Activity of S165A-cutinase*. To examine whether LC-cutinase* loses activity by the mutation of the catalytic serine residue, we constructed the mutant protein S165A-cutinase*, in which Ser165 is replaced by Ala. The far-UV CD spectrum of the mutant protein was similar to that of the wild-type protein (Fig. 2), suggesting that the mutation does not seriously affect the protein conformation. The specific activity of the mutant protein for pNP-butyrate (C4) was <0.03% of that of the wild-type protein at 30°C, indicating that LC-cutinase* almost fully loses activity by the mutation. S165A-cutinase* was released in the extracellular medium upon overproduction, as was the wild-type protein, indicating that the hydrolytic activity of LC-cutinase* is not necessary for leakage of the outer membrane.

Degradation of PCL. Poly(e-caprolactone) (PCL) is one of the synthetic aliphatic biodegradable polyesters (4). LC-cutinase* and Bc-Lip exhibited PCL-degrading activity, whereas Cr-Lip did not exhibit it (Fig. 3). Bc-Lip has been reported to degrade PCL (18). The amount of PCL degraded by LC-cutinase* at pH 8.0 and 50°C was 9.5 mg after incubation for 6 h. This amount increased in proportion to the incubation time. Thus, the specific PCL-degrading activity of LC-cutinase* at pH 8.0 and 50°C was determined to be 12 mg/h/mg of enzyme (2.7 mg/h/μkat of pNP-butyrate-degrading activity). These results indicate that LC-cutinase* can degrade a PCL film, but with much less efficiency than a PCL film. S165A-cutinase* did not degrade a PCL film, indicating that the hydrolytic activity of LC-cutinase* is responsible for degradation of PCL.

When the degradation products of PCL with LC-cutinase* were analyzed by reversed-phase high-pressure liquid chromatography, terephthalic acid (TPA) and mono(2-hydroxyethyl)terephthalate (MHET) were eluted from a column as a major and minor peak, respectively (see Fig. S5 in the supplemental material). The peak for bis(2-hydroxyethyl)terephthalate (BHET) was not detected. These results indicate that LC-cutinase* has an ability to completely degrade PCL to TPA and ethylene glycol.

Degradation of cutin. Cutin is composed of ω-hydroxy fatty acids, dihydroxyaliphatic acids, saturated and 12-monounsaturated 18-hydroxy 9,10-epoxy C18 acids, and saturated and 12-monounsaturated 9,10,18-trihydroxy C18 acids (23). Degradation of cutin with LC-cutinase* was analyzed by quantifying the carboxylic acids produced upon hydrolysis of ester bonds in cutin with 20 mM

**FIG 2** CD spectra. The far-UV spectra of LC-cutinase* (solid line) and S165A-cutinase* (dotted line) measured in 10 mM Tris-HCl (pH 7.0) containing 1 mM DTT are shown. These spectra were measured at 25°C as described in Materials and Methods.

**FIG 3** Degradation of PCL film by LC-cutinase*. A PCL film (20–30 mg) was incubated at 50°C for 6 h in 1 ml of 500 mM Tris-HCl (pH 8.0) containing 5 μg of LC-cutinase* (LCC), or 50 μg of B. cepacia lipase (Bc-Lip) or C. rugosa lipase (Cr-Lip), and its weight loss after incubation was determined. The experiment was carried out at least twice and the average values are shown, together with error bars.

**FIG 4** Degradation of PET film by LC-cutinase*. A PET film (20 to 25 mg) was incubated at 50°C for 24 h in 1 ml of 500 mM Tris-HCl (pH 8.0) containing 5 μg of LC-cutinase* (LCC), or 50 μg of Bc-Lip or Cr-Lip, and its weight loss after incubation was determined. The experiment was carried out at least twice, and the average values are shown together with error bars.
NaOH. The results are shown in Fig. 5. The amount of the carboxylic acids increased at 0.3 μmol/h (6 μmol/h/mg of enzyme) until it reached to 2.7 μmol at pH 8.0 and 50°C. This rate is comparable to that reported for T. fusca cutinase (4 μmol/h/mg of enzyme at pH 8 and 60°C) (9), suggesting that LC-cutinase* exhibits a comparable cutin-degrading activity as that of T. fusca cutinase. The carboxylic acids were not produced from cutin upon incubation with Bc-Lip, Cr-Lip, and S165A-cutinase*, suggesting that these enzymes do not have an ability to degrade cutin. However, neither the purity of the cutin used as a substrate nor the species of the hydrolytic products was determined. Further studies will be necessary to show that LC-cutinase* can really degrade cutin.

Possible application of LC-cutinase*. PET is widely used for industrial purposes, mainly as polyester fibers, beverage containers, and food packaging (10, 39). It is characterized by the high strength in chemical, physical, and mechanical properties. However, low hydrophilicity, poor wettability, and low moisture gain of polyester fibers cause various problems in manufacturing and consumer use. Hydrolysis and functionalization of polyester fibers with lipolytic and esterolytic enzymes are thought to be an effective way to improve surface properties of polyester fibers in an environmentally friendly manner (29, 39). Cutinase is one of these enzymes, and cutinases from F. solani f. sp. pisi (6, 10, 26, 36), F. oxysporum (25), and T. fusca (1, 13, 14, 24, 32) have been well studied regarding PET modification. However, the catalytic efficiencies of these enzymes are not sufficiently high to meet the requirements of the textile industry (10).

We showed here that a novel cutinase homolog isolated from leaf-branch compost with metagenomic approach exhibits a PET-degrading activity. It degrades a PET film at a rate of 12 mg/h/mg of enzyme (2.7 mg/h/μkat of pNP-butyrate-degrading activity). The degradation rate of a PET film with cutinase has been reported to be 0.05 mg/h/mg of enzyme for T. fusca cutinase at pH 7.0 and 55°C (24), 11.4 μg/h/μkat of pNP-butyrate-degrading activity for F. solani PBU-RU-5B cutinase at pH 7.0 and 25°C (26), 2.8 μg/h/μkat of pNP-butyrate-degrading activity for F. solani f. sp. pisi cutinase at pH 7.0 and 30°C (25), and 5.0 μg/h/μkat of pNP-butyrate-degrading activity for F. oxysporum cutinase at pH 7.0 and 30°C (25). Thus, the degradation rate of PET with LC-cutinase* is higher than the reported values for other cutinases by 230- to 970-fold. The catalytic efficiencies of F. solani f. sp. pisi (6) and T. fusca (32) cutinases have been successfully enhanced with protein engineering technology, but by only 5-fold. Therefore, LC-cutinase* might be potentially applicable for functionalization of PET fibers. In addition, detailed structural and functional analyses of LC-cutinase* will facilitate understanding of the mechanism by which LC-cutinase* hydrolyzes PET fibers and thereby lead to the development of a more efficient enzyme.

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