

Aerobic Denitrifying Bacteria That Produce Low Levels of Nitrous Oxide

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Received 25 November 2002/Accepted 17 March 2003

Most denitrifiers produce nitrous oxide (N₂O) instead of dinitrogen (N₂) under aerobic conditions. We isolated and characterized novel aerobic denitrifiers that produce low levels of N₂O under aerobic conditions. We monitored the denitrification activities of two of the isolates, strains TR2 and K50, in batch and continuous cultures. Both strains reduced nitrate (NO₃⁻) to N₂ at rates of 0.9 and 0.03 μmol min⁻¹ unit of optical density at 540 nm⁻¹ at dissolved oxygen (O₂) (DO) concentrations of 39 and 38 μmol liter⁻¹, respectively. At the same DO level, the typical denitrifier *Pseudomonas stutzeri* and the previously described aerobic denitrifier *Paracoccus denitrificans* did not produce N₂ but evolved more than 10-fold more N₂O than strains TR2 and K50 evolved. The isolates denitrified NO₃⁻ with concomitant consumption of O₂. These results indicated that strains TR2 and K50 are aerobic denitrifiers. These two isolates were taxonomically placed in the β subclass of the class Proteobacteria and were identified as *P. stutzeri* TR2 and *Pseudomonas* sp. strain K50. These strains should be useful for future investigations of the mechanisms of denitrifying bacteria that regulate N₂O emission, the single-stage process for nitrogen removal, and microbial N₂O emission into the ecosystem.

Nitrous oxide (N₂O) is a gaseous nitrogen oxide that is present at a concentration of about 350 ppb in the atmosphere. The concentration of this compound was maintained below 300 ppb in the global nitrogen cycle before the 20th century. However, recent reports suggest that the atmospheric concentration of N₂O is now increasing at a rate as high as 0.3% per year (1). N₂O has a 200- to 300-fold-stronger greenhouse effect than carbon dioxide (CO₂) and has the potential to destroy the ozone layer (17). Therefore, the N₂O balance is critical to the natural environment. The proposed sources of N₂O are chemical industries, burning fossil fuels, and biomass, as well as soil denitrification of nitrogenous compounds resulting from excess agricultural fertilizer (3, 6, 25). Another critical source of N₂O is wastewater treatment plants, in which considerable amounts of nitrogen pollutants removed from treated water are released into the atmosphere as N₂O, as well as dinitrogen (N₂).

Currently, nitrogen removal in wastewater treatment plants is essentially based on the activity of nitrifying and denitrifying microorganisms, both of which are inhabitants of activated sludge. Nitrifying bacteria aerobically oxidize ammonium contaminants to nitrite (NO₂⁻) and nitrate (NO₃⁻), which are then reduced by denitrifying bacteria to gaseous nitrogen forms such as N₂O and N₂. Efficient wastewater treatment relies on successively exposing water to aerobic and anaerobic conditions, since nitrification and denitrification are aerobic and anaerobic processes, respectively (4, 18). These properties represent a shortcoming of current systems since both denitrification and nitrification produce N₂O as a by-product in the absence of correctly regulated oxygen (O₂) concentrations (2,

20). For example, the denitrifying activity of most denitrifying bacteria is suppressed when anaerobiosis is insufficient and they cannot catalyze the final step of denitrification (reduction of N₂O to N₂) and produce N₂O (2, 8, 13, 19). Because complete removal of dissolved O₂ is difficult before the anaerobic denitrification that follows aerobic nitrification, current systems release considerable amounts of N₂O during denitrification.

To overcome this problem, novel aerobic denitrifying bacteria are required that could be used for constructing aerobic denitrifying processes. *Paracoccus denitrificans* (formerly *Thiosphaera pantotropha*) (2, 19, 20) is an aerobic denitrifier that has been isolated from activated sludge and it is the best-characterized such organism. This species reduces NO₃⁻ even in the presence of a saturating concentration of O₂. More recent surveys of aerobic denitrifiers have revealed some novel species, such as *Microvirgula aerodenitrificans* (16) and *Thaurea mechernichensis* (22); the former organism denitrifies as efficiently as *P. denitrificans* under aerobic conditions (15). Previously published results show that these denitrifiers remove NO₃⁻ quite efficiently from treated water or from culture medium. However, the previous reports did not address the effect of O₂ on the reduction of N₂O to N₂.

In the present paper we describe a novel method for screening and characterizing natural aerobic denitrifiers that produce N₂ gas by reducing NO₃⁻ under oxic conditions. The strains described here produce less N₂O under aerobic conditions than the previously described aerobic denitrifiers produce, indicating that they potentially could be used to construct an aerobic denitrifying system that emits low levels of N₂O.

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MATERIALS AND METHODS

Strains and media. *P. denitrificans* (*T. pantotropha*) ATCC 35512 originated from the American Type Culture Collection. *Pseudomonas stutzeri* ZoBell (=

RESULTS

ATCC 14405) was provided by W. G. Zumft. Other strains were isolated in this study. The following media were used: bromothymol blue (BTB) medium (0.1% L-asparagine, 0.1% KNO₃, 0.1% KH₂PO₄, 0.005% FeCl₂ · 6H₂O, 0.02% CaCl₂ · 2H₂O, 0.1% MgSO₄ · 7H₂O, 1 ml of BTB liter⁻¹ [1% in ethanol], 2% agar; pH 7.0 to 7.3), screening medium (SM) [0.284% sodium succinate, 10 mM NaNO₃, 0.136% KH₂PO₄, 0.027% (NH₄)₂SO₄, 0.1% yeast extract (Difco), 0.019% MgSO₄ · 7H₂O, 1 ml of a trace element solution (14) liter⁻¹; pH 7.2], denitrification medium (DM) (0.472% sodium succinate, 10 mM Na¹⁵NO₃, 0.15% KH₂PO₄, 0.042% Na₂HPO₄, 0.06% NH₄Cl, 0.5% Casamino Acids [Difco], 0.1% MgSO₄ · 7H₂O, 2 ml of a trace element solution [14] liter⁻¹; pH 7.2), artificial wastewater (AWW) (0.085% NaNO₃, 0.06% peptone, 0.04% bouillon extract, 0.01% urea, 0.003% NaCl, 0.01% KH₂PO₄, 0.0014% KCl, 0.002% MgSO₄ · 7H₂O, 0.00185% CaCl₂ · 2H₂O), and Luria-Bertani medium (1% tryptone, 0.5% yeast extract, 0.5% NaCl).

Screening of denitrifiers. Samples from rice ponds, domestic wastewater, and soil were transferred to 200 ml of SM in 500-ml Erlenmeyer flasks with cotton plugs and incubated at 30°C for 3 days. Fresh SM was inoculated with 5 ml of culture and incubated under the same conditions. These procedures were repeated three times. The resultant bacterial suspension was streaked on BTB medium plates with 8.5 g of sodium succinate liter⁻¹ and incubated at 30°C for 1 to 3 days. Soil samples were sometimes suspended in 0.9% NaCl and plated directly on BTB medium plates. Regardless, soil samples were incubated in DM or AWW instead of SM. Resulting blue colonies were isolated and screened as follows (second screening). The bacteria were transferred to 200 ml of Luria-Bertani medium with 10 mM NaNO₃ in a 500-ml Erlenmeyer flask. The flask was sealed with a butyl rubber stopper and rotary shaken at 120 rpm at 30°C (pre-culture). The atmospheric air in the headspace was not replaced, so the initial conditions were aerobic. A portion (50 ml) of the preculture was collected by centrifugation, washed twice with 0.9% NaCl, transferred to 100 ml of DM containing ¹⁵N (Na¹⁵NO₃) in a 500-ml Erlenmeyer flask, and incubated as described above. Aerobic denitrification by the bacteria was measured by determining time-dependent production of ¹⁵N₂ and the amount of residual O₂ in the headspace gas.

Batch culture. Batch cultures in flasks were incubated essentially under the conditions described above for the second screening. To investigate their effects on denitrification, various carbon sources were added to DM instead of succinate at a C/N ratio of 24.

Continuous culture. Precultures (100 ml) were transferred to 1-liter fermentation jars (BMJ-01; Able, Tokyo, Japan) containing 500 ml of DM or AWW with formate in which Na¹⁵NO₃ was substituted for Na¹⁴NO₃. Each culture was magnetically stirred at 300 rpm and 30°C (pH 7.4). Initially, 20% O₂ balanced with argon was added to the cultures at a constant flow rate of 20 ml/min. Exhaust gas that flowed into gas sampling tubes was collected after passing through a vapor condenser. The culture broth was aseptically withdrawn through sampling tubes, and then NO₃⁻, NO₂⁻, and biomass were analyzed. A continuous nutrient flow was started when cultures reached the logarithmic growth phase and was continued at a dilution rate of 0.14 h⁻¹. After 48 h, exhaust gas was analyzed in duplicate to confirm the steady state. The O₂ supply conditions were manipulated by changing the feed gas composition and/or the stirring speed. The cultures were constantly supplied with nutrients for over 48 h, which was equivalent to 6.7 replacements of the working volume to ensure a steady state.

Determination of the 16S rRNA gene sequences. Total bacterial DNA was purified as described previously (14). The genes encoding 16S rRNA were amplified by PCR by using total DNA (0.1 µg) as the template and primers AGA GTTTGATCCTGGCTCAG and GGTTACCTGTACGACTT (26). Genes were amplified by 30 cycles of 94°C for 20 s, 50°C for 1 min, and 72°C for 1.5 min, followed by extension at 72°C for 10 min.

Analytical methods. We analyzed gaseous O₂ and N₂O by gas chromatography (GC) (23). Isotope-labeled N₂ gas was measured by isotope mass spectroscopy (Delta plus, Finnigan MAT) as described previously (24), and NO₃⁻ and NO₂⁻ were measured colorimetrically (12).

Enzyme assays. Nitrate reductase (Nar) activity was assayed as described previously (7) by using methylviologen-dithionite as the electron donor. Nitrite reductase (Nir) activity was assayed by using NADH-phenazine methosulfate as the electron donor by determining the amount of NO produced by the P450nor-trap method (7). Nitric oxide reductase (Nor) activity was determined by determining NADH-dependent N₂O formation by GC as described previously (11). Nitrous oxide reductase activity was assayed as described by Kshirsagar et al. (9).

Nucleotide sequence accession numbers. The nucleotide sequence data have been deposited in the DDBJ/EMBL/GenBank nucleotide sequence databases under accession numbers AB096261 (strain TR2) and AB096260 (strain K50).

Isolation of novel aerobic denitrifiers. We developed plate assays to isolate bacterial strains that denitrify under aerobic conditions. The method is based on the changes in the pH of a medium due to NO₃⁻ depletion by denitrification. The plates contained KNO₃ and the pH indicator BTB. The pH of the medium was initially adjusted to between 7.0 and 7.3. Plates inoculated with a bacterial suspension were incubated for several days at the appropriate temperature, and then bacterial NO₃⁻ consumption was monitored by examining blue colonies and/or halos due to the increasing pH of the medium. Positive strains obtained from this initial screening analysis were screened further by culturing them in flasks containing DM under initially aerobic conditions. Denitrification was periodically monitored by measuring N₂O and ¹⁵N₂ in the gas phase by GC and GC-mass spectrometry, respectively. Most strains completely consumed O₂ before they produced N₂. However, some strains produced significant amounts of N₂ even in the presence of 3% O₂ in the gas phase of the flasks. Among these, strains that produced the least N₂O (aerobic denitrifiers emitting low N₂O levels) were selected.

In a typical experiment we screened 97 bacterial colonies, and we found 29 strains which were positive in BTB plate assays and 9 strains that produced N₂ even in the presence of 3% O₂. Six of the nine strains evolved less than 1 µmol of N₂O flask⁻¹ under the same culture conditions. The most prominent aerobic (O₂-tolerant) N₂ producers (strains TR2 and K50) from several screening analyses were analyzed as follows.

Properties of denitrifiers in batch culture. The time-dependent denitrification by strain TR2 or K50 was compared with that of the typical denitrifiers *P. stutzeri* ZoBell and *P. denitrificans* ATCC 35512 during batch culture in flasks with a headspace that was initially filled with air to create aerobic conditions. After the incubation was started, the O₂ concentration in the gas phase was gradually decreased, and N₂ evolved (Fig. 1). The rates of N₂ production were significantly different for the different cultures. Initiation of N₂ production by *P. stutzeri* ZoBell required a lag period of 2.5 h, whereas the other three strains evolved N₂ without a lag phase, indicating that N₂ production by *P. stutzeri* ZoBell is more sensitive to O₂ than N₂ production by the other strains is (Fig. 1A). The aerobic denitrifier *P. denitrificans* produced N₂ without a lag but concomitantly produced a considerable amount of N₂O (Fig. 1B). Strains TR2 and K50 produced N₂ without a lag period, and little N₂O was detected during the entire incubation period. These results demonstrated that strains TR2 and K50 should be aerobic denitrifiers that produce low levels of N₂O even under aerobic conditions. The NO₃⁻ initially added (1 mmol) was completely consumed by these cultures (data not shown). The amount of consumed N atoms recovered in N₂ was greatest with strain K50, in which the 0.27 mmol of N₂ obtained corresponded to 54% recovery (Fig. 1C). The final level of recovery with strain TR2 was 28% (Fig. 1D).

Denitrification in continuous culture. We further investigated the effects of aeration on N₂ and N₂O production by the isolated bacteria. Table 1 shows the steady-state characteristics observed with continuous cultures of the strains isolated and of the typical denitrifiers. Changing the composition of the aerating gas and agitation speed allowed us to establish anoxic

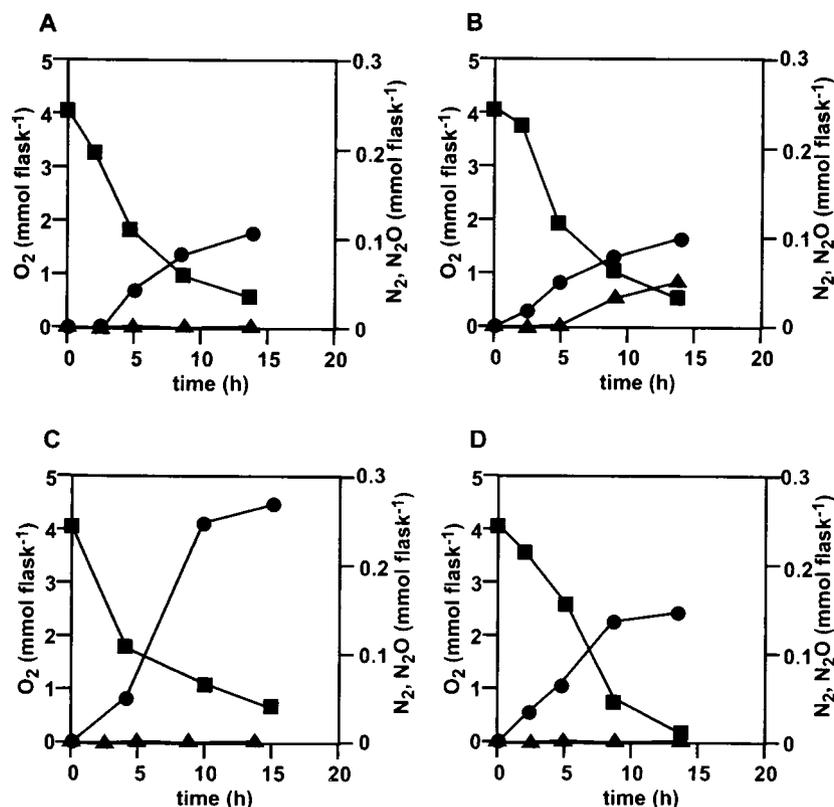


FIG. 1. Denitrification by new isolates and control denitrifiers in initially aerobic batch cultures. The strains were cultured in flasks containing DM, and the gas phase was monitored periodically as described in Materials and Methods. (A) *P. stutzeri* ZoBell; (B) *P. denitrificans* ATCC 35512; (C) strain K50; (D) strain TR2. Symbols: ●, N₂; ▲, N₂O; ■, O₂. Typical results from more than three independent experiments are shown.

(dissolved oxygen [DO] concentration, 0 $\mu\text{mol liter}^{-1}$), hypoxic (DO concentration, 5.3 to 9.4 $\mu\text{mol liter}^{-1}$), and oxic (DO concentration, 28 to 39 $\mu\text{mol liter}^{-1}$) aeration conditions. Most of the cultures produced N₂ under anoxic and hypoxic conditions; the only exception was a *P. denitrificans* culture in which there was too little growth to detect N₂ produced under anoxic conditions. All of the strains produced N₂ at a higher

rate under hypoxic conditions than under anoxic conditions (data not shown). However, under anoxic conditions the rate of N₂ production was higher than that under hypoxic conditions, as for other bacterial denitrifiers (2, 4), when the rates were expressed on the basis of cell mass (optical density at 540 nm [OD₅₄₀]). This is because the cell mass was significantly larger under hypoxic conditions than under anoxic conditions.

TABLE 1. Steady-state characteristics of continuous cultures of bacterial strains^a

Strain	Conditions ^b	DO concn ($\mu\text{mol liter}^{-1}$)	OD ₅₄₀	NO ₃ ⁻ consumption ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)	N ₂ O production ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)	N ₂ production ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)
TR2	A	0	0.5	24	9.9×10^{-5}	6.8
	H	5.3	1.3	7.3	9.7×10^{-4}	2.8
	O	39	2.4	3.5	1.6×10^{-4}	0.9
K50	A	0	0.4	30	9.5×10^{-3}	6.5
	H	6.3	1.7	6.8	1.6×10^{-3}	1.8
	O	38	3.7	0	6.2×10^{-4}	0.03
<i>P. stutzeri</i> ZoBell	A	0	0.5	24	1.5×10^{-2}	2.9
	H	7.8	1.3	8.8	6.7×10^{-3}	1.5
	O	38	1.0	8.4	6.2×10^{-3}	0.0
<i>P. denitrificans</i> ATCC 35512	A	0	0.2	60	9.5×10^{-3}	0.0
	H	9.4	3.2	2.7	4.1×10^{-3}	1.4
	O	28	6.7	0.4	2.8×10^{-2}	0.0

^a DM was used for the cultures. The dilution rate was 0.14 h⁻¹. The data are averages of two independent experiments.

^b A, anoxic (the influent gas was 100% Ar, and the agitation rate was 300 rpm); H, hypoxic (the influent gas was 20% O₂-80% Ar, and the agitation rate was 300 rpm); O, oxic (the influent gas was 20% O₂-80% Ar, and the agitation rate was 430 rpm).

TABLE 2. Carbon source utilization of strains TR2 and K50^a

Carbon source	Reaction rate ($\mu\text{mol h}^{-1}$)					
	Strain TR2			Strain K50		
	N ₂	N ₂ O	O ₂	N ₂	N ₂ O	O ₂
Glycerol	8.8	1.3	230	4.8	0.05	140
Glucose	8.8	6.5	230	15.0	ND ^b	390
Succinate	36.3	ND	380	42.5	ND	410
Citrate	17.5	2.1	310	16.3	0.23	310
Acetate	13.8	2.6	250	28.8	ND	410
Ethanol	2.5	8.3	263	28.8	ND	450
Methanol	0.50	1.6	230	0.04	0.73	150
Formate	0.16	12.0	200	ND	1.0	380

^a The values are the initial reaction rates for N₂ and N₂O production and O₂ consumption during the first 8 h of flask culture. N₂, N₂O, and O₂ concentrations in the gas phase were determined by GC. The initial amounts of NO₃⁻ and O₂ were 1 and 4 mmol, respectively. The data are averages of two independent experiments.

^b ND, not detected.

The value per cell for strain TR2 (2.8 μmol of N₂ min⁻¹ unit of OD₅₄₀⁻¹) was double that for *P. denitrificans* (1.4 μmol of N₂ min⁻¹ unit of OD₅₄₀⁻¹) under hypoxic conditions. When challenged with oxic conditions, *P. stutzeri* ZoBell and *P. denitrificans* produced no N₂ and considerable amounts of N₂O, indicating that O₂ inhibited the reduction of N₂O to N₂ under these conditions. By contrast, strains TR2 and K50 could still produce N₂ under oxic conditions (Table 1). The activity of strain TR2 (0.9 μmol of N₂ min⁻¹ unit of OD₅₄₀⁻¹) was as much as 32% of the activity under hypoxic conditions. Both TR2 and K50 produced less N₂O than the other strains produced under the corresponding aeration conditions. TR2 produced less N₂O than K50 produced, indicating that strain TR2 reduces N₂O to N₂ more efficiently than K50 reduces N₂O to N₂. Our results showed that *P. denitrificans*, a typical aerobic denitrifier, produced N₂O at a much higher rate than the other strains produced N₂O, indicating that this organism produces N₂O instead of N₂ under oxic conditions. These findings indicate that the strains isolated in the present study are aerobic denitrifiers that produce low levels of N₂O.

Electron donor specificity. We investigated the electron donor specificity of denitrification by strains TR2 and K50 in flask cultures under initially aerobic conditions (Fig. 1). Table 2 shows that both strains consumed O₂ and produced N₂ and/or N₂O with all carbon sources tested, indicating that they can use a variety of carbon sources as electron donors for O₂ respiration and denitrification. Of the carbon sources tested, succi-

nate supported O₂ respiration and denitrification most efficiently in TR2. Strain K50 utilized O₂ efficiently in the presence of ethanol and acetate in addition to succinate. Although the N₂ production rates were lower and more N₂O evolved than the N₂O that evolved with the other carbon sources, C₁ compounds, such as methanol and formate, could support denitrification by both bacteria. Strain K50 evolved little N₂O irrespective of the electron donor.

Identification of the strains. Strains TR2 and K50 were both gram-negative rods with catalase and oxidase activities. They also metabolized glucose to organic acid, indicating that they are pseudomonads. The nucleotide sequences of their 16S rRNA genes and a phylogenetic analysis revealed that these strains are members of the genus *Pseudomonas* in the β subclass the class *Proteobacteria*. The levels of identity of the sequences of strains TR2 and K50 were greatest with the sequences of *P. stutzeri* (99%) and *Pseudomonas mendocina* NCIB 10541 (99%), respectively. The phenotype of strain TR2 was characteristic of *P. stutzeri*; namely, there were no fluorescent pigments, organic growth factors were not required, and the organism was gelatinase negative and amylase positive. The ability to grow at 41°C is also a phenotypic criterion that distinguishes *P. stutzeri* strains from other *Pseudomonas* spp. Thus, we identified strain TR2 as *P. stutzeri* and designated it *P. stutzeri* TR2. Strain K50 accumulated poly β -hydroxybutyric acid, which *P. mendocina* does not do. Therefore, we could not identify this strain and designated it *Pseudomonas* sp. strain K50.

Performance in AWW system. We estimated the denitrification by continuous cultures of *P. stutzeri* TR2 in organic wastewater. We added formate to conventional wastewater as an additional electron donor assuming that it should provide selective pressure for survival that should allow *P. stutzeri* TR2 to become the dominant species in the culture. Table 3 shows that neither *P. stutzeri* TR2 nor the control strain *P. stutzeri* ZoBell evolved N₂O at any DO level. TR2 more efficiently reduced NO₃⁻ and produced more N₂ than *P. stutzeri* ZoBell. When the DO concentration was increased to 113 $\mu\text{mol liter}^{-1}$, *P. stutzeri* ZoBell produced neither N₂ nor N₂O. This is in contrast to *P. stutzeri* TR2, which produced a significant amount of N₂ even under highly aerobic conditions (DO concentration, 141 $\mu\text{mol liter}^{-1}$). The level of recovery of N atoms in N₂ in an aerobic culture of *P. stutzeri* TR2 was 14% of the consumed NO₃⁻. Other N atoms of the consumed NO₃⁻ should have been incorporated into the biomass. These results indicate that the denitrifying system of *P. stutzeri* TR2 is more resistant to

TABLE 3. Steady-state characteristics of strains in AWW supplemented with formate^a

Strain	DO concn ($\mu\text{mol liter}^{-1}$)	OD ₅₄₀	NO ₃ ⁻ consumption ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)	NO ₂ ⁻ production ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)	N ₂ production ($\mu\text{mol min}^{-1}$ unit of OD ₅₄₀ ⁻¹)
<i>P. stutzeri</i> TR2	3	0.22	51.4	7.7	5.5
	72	0.50	23.0	0.18	5.6
	141	0.62	18.2	0.10	1.3
<i>P. stutzeri</i> ZoBell	3	0.45	23.1	0.89	1.3
	113	0.50	22.4	0.14	ND
	160	0.49	22.7	0.08	ND

^a The dilution rate was 0.14 h⁻¹. The data are typical results of three independent experiments. No N₂O production was detected for any strain.

O₂ than the denitrifying system of *P. stutzeri* ZoBell is and that the TR2 strain produced N₂ under aerobic conditions in AWW.

Enzyme activities. The activities of respiratory NO₃ reductase, NO₂⁻ reductase, NO reductase, and N₂O reductase in crude extracts prepared from denitrifying cells of *P. stutzeri* TR2 were determined, and the results obtained with 3-day cultures are shown in Fig. 1. The specific activities of these enzymes were 4,400, 63, 6.8, and 64 nmol of product min⁻¹ mg of protein⁻¹, respectively, indicating that the activity observed should have been responsible for denitrification by *P. stutzeri* TR2.

DISCUSSION

Although N₂O production by denitrifying bacteria under insufficient anaerobic conditions is an established phenomenon that causes serious global warming, in most studies of aerobic denitrifiers the workers have described only NO₃⁻ and NO₂⁻ consumption under aerobic conditions, and studies in which the workers focused on denitrified gas (N₂O and N₂) have been limited. This is probably due to difficulties in detecting denitrified N₂ under aerobic conditions; under such conditions contamination with atmospheric N₂ interferes with precise quantitation of denitrified N₂. Here, we used heavy-isotope-labeled NO₃⁻ (¹⁵NO₃⁻) and highly sensitive mass spectrometry to screen and isolate denitrifying bacteria emitting low levels of N₂O under aerobic conditions. To our knowledge, we are the first researchers to isolate and characterize denitrifiers that produce less N₂O than other bacteria produce. Our screening method should be applicable to isolation of aerobic denitrifying bacteria that exhibit such properties.

Our results indicated that both of the denitrifiers isolated, *P. stutzeri* TR2 and *Pseudomonas* sp. strain K50, produced N₂ even under oxic conditions (DO concentration, 38 to 39 μmol liter⁻¹) (Table 1), conditions under which other typical aerobic denitrifiers could not produce N₂, indicating that the novel strains are O₂ resistant, aerobic denitrifiers. Physiologically, O₂ is the best electron acceptor for supporting growth, and it receives electrons through the respiratory chain. Most denitrifying microorganisms perform O₂ respiration and repress the denitrification mechanism when O₂ is available. Therefore, the physiological significance of denitrification under oxic conditions (aerobic denitrification) is quite intriguing. Aerobic denitrification has also been identified in *P. denitrificans* (19, 20), as well as in the novel species *M. aerodenitrificans* (15) and *T. mechernichensis* (22). The strains identified here, together with these organisms, should be useful for future investigations of the mechanisms of aerobic denitrification by bacteria and the role of aerobic denitrification in the ecosystem.

We found that all of the bacteria tested emitted considerably more N₂O under continuously aerobic conditions than under anaerobic conditions (Table 1). This is consistent with the observation that in most denitrifiers, the activity of N₂O reductase is inactivated by O₂, which also represses expression of the encoding gene and N₂ production (4, 8). We also found that both *P. stutzeri* TR2 and *Pseudomonas* sp. strain K50 produced less N₂O and more N₂ under our conditions than the other denitrifiers produced (Fig. 1 and Table 1). Furthermore, our results demonstrated that the N₂- and N₂O- producing

activities of the strains were dependent on the culture. For example, carbon sources seemed to affect denitrification to various degrees depending on the strain (Table 2). Factors other than carbon sources can also affect aerobic denitrification. In the presence of formate as an electron donor, *P. stutzeri* TR2 produced N₂O in batch culture (Table 2), whereas little N₂O was detected in continuous culture (Table 3). This may have been due to differences in conditions between the batch and continuous cultures or to the compositions of the media. The mechanism with which the isolates suppressed N₂O production remains to be studied. However, the results suggest that aerobic denitrification that produces less N₂O should be a complicated system that is regulated by various factors, including the O₂ supply and the composition of the medium. It is interesting that *P. stutzeri* TR2 and ZoBell have different denitrifying properties, especially for N₂O production under oxic conditions (Table 1) and for N₂ production in response to sudden exposure to O₂ (Fig. 1), despite their close phylogenetic relationship. A comparison of the genes of these strains should reveal the critical gene(s) required for aerobic denitrification and for N₂O suppression.

The novel bacteria isolated in the present study denitrified under conditions that mimicked those in wastewater treatment plants (Table 3). This finding suggests that the isolates could be maintained in the mixed population of microorganisms found in sludge as is observed with *P. denitrificans*, which reduces NO₃⁻ more efficiently than a control reactor without the strain when it is mixed with activated sludge (9). Sometimes, a C₁ carbon source, such as methanol or formate, has been used to manipulate the dominant bacterial species in sludge (5, 21). Our novel strains should be useful for constructing new aerobic denitrification processes that do not produce N₂O, since these strains could be selected as the dominant organisms by using C₁ compounds as the electron donors.

ACKNOWLEDGMENTS

We thank Kota Hatayama (University of Tsukuba) for determining the nucleotide sequence of the 16S rRNA gene. We also thank Norma Foster for critical reading of the manuscript.

This study was supported by PROBRAIN (Program for Promotion of Basic Research Activities for Innovative Biosciences) and by a grant-in-aid for scientific research from the Ministry of Education, Science, Culture and Sports of Japan.

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