Microbial species diversity, community dynamics, and metabolite kinetics of water kefir fermentation

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Running title

Kinetic analysis of water kefir fermentation

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Abstract

Water kefir is a sour, alcoholic, and fruity fermented beverage, of which the fermentation is started with water kefir grains. These water kefir grains consist of polysaccharide and contain the micro-organisms responsible for the water kefir fermentation. In this work, a water kefir fermentation process was followed as a function of time during 192 h to unravel the community dynamics, the species diversity, and the kinetics of substrate consumption and metabolite production. The majority of the water kefir ecosystem was found to be present on the water kefir grains. The most important microbial species present were *Lactobacillus casei/paracasei, Lactobacillus harbinensis, Lactobacillus hilgardii, Bifidobacterium psychraerophilum/crudilactis, Saccharomyces cerevisiae*, and *Dekkera bruxellensis*. This microbial species diversity was similar in the water kefir liquor and on the water kefir grains, and remained stable during the whole fermentation process. The major substrate, sucrose, was completely converted after 24 h of fermentation, which coincided with the production of the major part of the water kefir grain polysaccharide. The main metabolites of the fermentation were ethanol and lactic acid. Glycerol, acetic acid, and mannitol were produced in low concentrations. The major part of these metabolites was produced during the first 72 h of fermentation, during which the pH decreased from 4.26 to 3.45. The most prevailing volatile aroma compounds were ethyl acetate, isoamyl acetate, ethyl hexanoate, ethyl octanoate, and ethyl decanoate, which might be of impact on the aroma of the end product.
Introduction

Water kefir is a fermented beverage that is made by adding water kefir grains, these are polysaccharide grains that serve as the inoculum, to a mixture of water, sugar (sucrose), dried figs, and possibly other ingredients such as lemon, depending on the recipe (1-5). After 2 to 4 days of anaerobic incubation at room temperature, a sparkling, yellowish fermented beverage is obtained that has a fruity, acidic, slightly sweet, and slightly alcoholic taste and aroma.

Water kefir is available worldwide but it is still unknown what the real origin of the water kefir grains is. Once it has been postulated that the polysaccharide grains originate from the leaves of the *Opuntia* cactus fig plant (6). Besides the use of the name water kefir grains in western Europe, other names are also in use for this fermented beverage inoculum, depending on the geographic location, such as ‘ginger beer plants’, ‘Tibicos’, ‘Tibi grains’, ‘California bees’, ‘African bees’, ‘ale nuts’, ‘balm of Gilead’, ‘Bèbées’, ‘Japanese beer seeds’, and ‘sugary kefir grains’ (2, 5-9).

Currently, research on water kefir is still very limited and most of the scientific information available deals with its species diversity (2, 4, 5, 8-16). Also, the chemical and structural composition of the water kefir grain polysaccharide has been studied (6, 7, 10-12, 15, 17-19).

To date, it is known that the microbial species diversity of water kefir consists of a stable consortium of mainly lactic acid bacteria, yeasts and acetic acid bacteria, as shown by both culture-dependent and culture-independent techniques (2, 4, 19, 20). Recently, also *Bifidobacterium psychraerophilum/crudilactis* was found in water kefir via culture-dependent and culture-independent techniques (2). It became however clear that different water kefirs display different species diversities. Hence, a systematic approach for the study of the microbiology of water kefir fermentation is necessary. Further, it is known that the water kefir grain polysaccharide consists of dextran, an α-(1→6)-linked glucose polymer, produced by...
certain *Lactobacillus* and/or *Leuconostoc* species (5, 7, 11, 17-19). However, until now no thorough metabolite analysis has been performed on a water kefir fermentation process. The aim of the present study was to gain a deeper understanding of the water kefir fermentation process, in particular its microbial species diversity, community dynamics, substrate consumption profile, and metabolite production course.

**Materials and methods**

**Preparation of the inoculum**

To prepare an inoculum, approximately 100 g of water kefir grains were obtained from a private person, who cultivated water kefir at home (Ghent, Belgium). To obtain the necessary amount of kefir grains (± 600 g) to start an actual water kefir fermentation process, a series of consecutive pre-fermentations was performed in a common water kefir cultivation medium that was used at household level. Therefore, for every 15 g of water kefir grains, 6 g of unrefined cane sugar (Candico Bio, Merksem, Belgium), 85 ml of tap water, and 5 g of dried figs (King Brand, Naziili, Turkey) were added. These pre-fermentations were performed in Schott bottles (1, 2, and 5 l consecutively) with a water lock from polytetrafluoroethylene (PTFE) that were incubated in a water bath at 21 °C. A backslopping practice was applied every 72 h. Therefore, the water kefir grains were separated from the fermenting medium by sieving and recultivated in fresh medium under the same conditions. After each backslopping the water kefir grain mass increased and this practice was applied until > 600 g of water kefir grains were obtained.

**Water kefir fermentation process**

The pre-cultivated water kefir grains served as inoculum for the water kefir fermentations (in triplicate) in 100-ml Schott bottles (three times twelve bottles), equipped with water lock.
Each bottle fermentation was started at the same time with 15 g of water kefir grains and 85 ml of sterile water kefir simulation medium. The water kefir simulation medium was prepared with 6 g of unrefined cane sugar (Candico Bio), 65 ml of tap water, and 20 ml of fig extract. Fig extract was prepared by adding 20 ml of distilled water to 5 g of dried figs, after which this suspension was mixed finely and centrifuged (7200 x g, 20 min, 4°C). The supernatant was filtered through a coffee filter to obtain the final fig extract. The fermentation bottles were incubated in a water bath at 21 °C. After closure of the bottles at the start of the fermentations, as well as before each sampling, the contents were homogenized by mildly turning the bottles. After 0, 3, 6, 12, 18, 24, 36, 48, 72, 96, 144, and 192 h of fermentation, the contents of three Schott bottles (representing the three independent fermentations) were analyzed. The results at each sampling point are presented as the mean ± standard deviation.

Water kefir grain mass and pH measurements

After opening the bottles, their contents were sieved to separate the water kefir grains from the water kefir liquor. The water kefir grains were washed with 200 ml of sterile saline [8.5 g l\(^{-1}\) of NaCl (Merck KGaA, Darmstadt, Germany)] and the total water kefir grain mass (wet mass) was weighed. Then, approximately 5 g of these water kefir grains were transferred into an aluminum recipient and dried at 105 °C for 48 h to determine the dry mass of the water kefir grains. The pH of the water kefir liquor was measured with a SenTix 41 glass electrode (WTW GmbH, Weilheim, Germany).

Microbial enumerations

To enumerate micro-organisms in the water kefir liquor, the water kefir liquor obtained as described above was used to prepare appropriate decimal dilutions in sterile saline. For the enumerations of the micro-organisms on the water kefir grains, 5.0 g of washed water kefir grains obtained as described above were brought into a sterile stomacher plastic bag. These
grains were crushed by rolling a glass bottle over the outside of the bag, after which 45 ml of sterile maximum recovery diluent [8.5 g l\(^{-1}\) of NaCl (Merck) and 1 g l\(^{-1}\) of bacteriological peptone (Oxoid, Basingstoke, Hampshire, UK)] was added. This mixture was homogenized for 15 min at high speed in a Stomacher 400 (Seward, Worthington, UK). The resulting water kefir grain suspension was used to prepare appropriate decimal dilutions in sterile saline.

The diluted suspensions were plated on de Man, Rogosa and Sharpe (MRS) agar medium, modified deoxycholate-mannitol-sorbitol (mDMS) agar medium, and yeast extract-glucose (YG) agar medium to enumerate presumptive lactic acid bacteria, acetic acid bacteria, and yeasts, respectively (21, 22). In addition, cycloheximide (final concentration of 0.1 g l\(^{-1}\); Sigma-Aldrich, Saint Louis, MO) was added to the MRS and mDMS agar media to inhibit fungal growth, whereas chloramphenicol (final concentration of 0.1 g l\(^{-1}\); Sigma-Aldrich) was added to YG agar medium to inhibit bacterial growth. Further, the water kefir liquor and water kefir grain suspensions at 0 and 72 h of fermentation were plated on kanamycin-aesculin-azide (KAA) agar medium (Oxoid) and violet-red-bile-glucose (VRBG) agar medium (Oxoid) to enumerate presumptive enterococci plus streptococci and Enterobacteriaceae, respectively. MRS, mDMS, and YG agar media were incubated at 30 °C for 2 to 4 days; KAA and VRBG agar media were incubated at 42 °C for 24 h. Colony enumerations were expressed as log colony forming units (cfu) per ml of water kefir liquor or per g of water kefir grains.

**Culture-dependent species diversity analysis**

The culture-dependent species diversity analysis of lactic acid bacteria (based on MRS agar isolates), acetic acid bacteria (based on mDMS agar isolates), and yeasts (based on YG agar isolates) in the water kefir liquor and on the water kefir grains was assessed after 0, 24, 48, 72, and 192 h of fermentation. This was done by randomly picking up colonies (10 to 20 % of...
the total colony count) from the agar plates with 30 to 300 colonies. The bacterial and yeast colonies were subcultivated in MRS medium (30 °C, 24 h) and yeast extract-glucose-peptone (YGP) medium (30 °C, 24 h), respectively. These cultures were supplemented with glycerol (final concentration of 25 %, v v⁻¹) and stored at –80 °C.

To harvest cells for DNA extraction, 2 ml of overnight cultures were centrifuged (21,000 x g, 5 min, 4 °C) and the supernatant was discarded. The DNA of the bacterial and yeast cultures was extracted and purified with the Nucleospin® 96 tissue kit (Macherey-Nagel GmbH, Düren, Germany), according to the instructions of the manufacturer. Bacterial cultures were treated with mutanolysin (Sigma-Aldrich) and lysozyme (VWR), and the yeast cultures were treated with lyticase (Sigma-Aldrich). Bacterial DNA was diluted to approximately 50 ng µl⁻¹ and was subjected to (GTG)₅-PCR fingerprinting as described before (23). Yeast DNA was diluted to approximately 20 ng µl⁻¹ and was subjected to M13-PCR fingerprinting as described before (24). The fingerprint patterns obtained were clustered into similarity trees [based on the Pearson correlation coefficient and the unweighted pair group method with arithmetic average (UPGMA) algorithm] with Bionumerics 5.10 software (Applied maths, Sint-Martens-Latem, Belgium). Identification of the clusters was performed by the identification of several representatives within each cluster.

For identification of the bacteria, genomic DNA of the selected isolates was subjected to a PCR assay to amplify the 16S rRNA gene (1.5 kb) with primer pair pA/pH (25). For the identification of the yeast isolates, the primer pair ITS1/ITS4 (26) was used to amplify the internal transcribed spacer (ITS) region (variable length) and the primer pair LR0R/LR3 (27) was used to amplify a part of the large subunit (LSU) rRNA gene (0.6 kb). These PCR amplicons were purified with the Wizard Plus SV Minipreps DNA Purification System (Promega, Madison, WI) and sequenced in a commercial facility (Macrogen, Amsterdam, The
Culture-independent species diversity analysis

For the culture-independent analyses of the micro-organisms in the water kefir liquor and on the water kefir grains, 40 ml of sieved water kefir liquor and 10 ml of a water kefir grain suspension were centrifuged (7,200 x g, 20 min, 4 °C). These pellets were resuspended in 2 ml of TES buffer [6.7 % (m v⁻¹) sucrose (Merck), 50 mM Tris-base (Merck), 1 mM EDTA (Sigma-Aldrich), pH = 8.0] and centrifuged (21,000 x g, 20 min, 4 °C). The resulting pellets were used for DNA extraction. Therefore, the pellets were treated with 200 U of lyticase (Sigma-Aldrich) in 600 µl of sorbitol buffer (30 °C, 60 min) and centrifuged (21,000 x g, 5 min, 4 °C), and further processed as described previously (29). The DNA obtained was further purified with the Nucleospin® food kit (Macherey-Nagel), according to the instructions of the manufacturer. The purified DNA was diluted to approximately 50 ng µl⁻¹ before PCR assays were performed. To assess the bacterial diversity, the universal 16S rRNA primer pair 357f-GC/518r (V3) was used (30). The group-specific primer pair LAC1/LAC2-GC (LAC) was used for lactic acid bacteria (31) and the genus-specific primer pair bif164f/bif662r-GC (Bif) was used for bifidobacteria (32). A GC clamp was attached to one primer of each primer pair, as indicated. The PCR amplicons were separated in a 6 % (v v⁻¹) polyacrylamide gel via denaturing gradient gel electrophoresis, as described before (21, 31). The denaturing gradients of the gels were, from top to bottom, 45–60 % for the V3 and the Yeast primer pairs, 40–55 % for the LAC primer pair, and 45–55 % for the Bif primer pair. Gel processing and DNA band sequencing were performed as described previously (21, 31). Identification of the DNA band sequences was performed as described above.
Substrate consumption and metabolite production

For the measurement of the substrates, the metabolites, and the aroma compounds, sieved water kefir liquor was centrifuged (7,200 x g, 20 min, 4 °C) to obtain cell-free supernatant. Carbohydrate concentrations were measured through high-performance anion exchange chromatography with pulsed amperometric detection (HPAEC-PAD) as described before (34). Therefore, 50 µl of cell-free supernatant was added to 950 µl of ultrapure water. From this dilution, 50 µl was added to 950 µl of deproteinization solution [500 µl of acetonitrile (Sigma-Aldrich), 449.5 µl of ultrapure water, and 0.5 µl of 50 g l⁻¹ of rhamnose (internal standard; Sigma-Aldrich)].

Glycerol and mannitol concentrations were measured through HPAEC-PAD as described before (35). Therefore, 100 µl of cell-free supernatant was added to 400 µl of ultrapure water. From this dilution, 100 µl was added to 900 µl of deproteinization solution. Quantification was performed with an external standard curve with standards prepared in the same way as the samples.

Lactic acid and acetic acid concentrations were measured through high-performance liquid chromatography with refractive index detection, as described previously (36). Therefore, 800 µl of cell-free supernatant was added to 200 µl of 50 % (m/v) trichloroacetic acid. Quantification was performed with an external calibration curve with standards prepared in the same way as the samples.

Ethanol concentrations were measured through gas chromatography (GC) with flame ionization detection, as described before (37). Therefore, 100 µl of cell-free supernatant was added to 1100 µl of deproteinization solution [720 µl of acetonitrile (Sigma-Aldrich), 367.7 µl of ultrapure water, 12 µl of formate (VWR), and 0.3 µl of 1-butanol (internal standard);
Quantification was performed with an external standard curve with standards prepared in the same way as the samples. After deproteinization, all samples were vortexed, centrifuged (21,000 x g, 20 min), filtered (0.2-µm Whatman™ filters; GE Healthcare Life Sciences, Bucks, United Kingdom), and injected onto the column. Volatile aroma compounds in the water kefir liquor were measured through static headspace gas chromatography coupled to mass spectrometry (SH-GC-MS). Therefore, 5 ml of cell-free supernatant was brought into a 20-ml glass headspace vial (Gerstel GmbH & Co. KG, Mülheim-an-der-Ruhr, Germany) and closed with a magnetic screw cap (18 mm) with a silicon/PTFE septum (Gerstel GmbH & Co. KG). Before analysis, 1.5 g of NaCl and 100 µl of internal standard solution [0.5 ml l⁻¹ of 4-methyl-2-pentanol (Sigma-Aldrich)] were added. The headspace vials were equilibrated at 40 °C for 30 min at 400 rpm in a MPS2 Gerstel autosampler (Gerstel GmbH & Co. KG), and further analyzed as described before (23). The compounds were identified by comparison of the mass spectra with library data (NIST 08 database, http://www.nist.gov). For quantification, an external calibration curve was constructed to get an indication of the concentrations present. The standards were prepared in ultrapure water and analyzed in the same way as the samples. All volatile aroma compounds found in the water kefir fermentation samples taken after 72 h of fermentation were compared with their threshold values as reported in the literature.

**Carbon recovery**

At each sampling time, the carbon recovery was calculated as the total amount of carbon at that sampling time point divided by the total amount of carbon at 0 h. The total amount of recovered carbon was calculated as the sum of the amount of carbon in the water kefir liquor, in the water kefir grains, and produced as carbon dioxide. For these calculations, the...
measurements of the water kefir grain mass and water kefir grain dry mass, and the sucrose, glucose, fructose, ethanol, lactic acid, glycerol, acetic acid, and mannitol concentrations were used. It was assumed that the water kefir grain density was 1 g cm\(^{-3}\), that the dried water kefir grain mass consisted of pure polysaccharide, and that the ethanol and acetic acid present in the water kefir grain matrix were evaporated during the dry mass determinations. The ethanol and acetic acid concentrations in the water kefir grain matrix were assumed to be the same as those in the water kefir liquor.

Results

Water kefir grain mass and pH measurements

A water kefir fermentation process, inoculated with water kefir grains that were obtained through a series of pre-fermentations, was carried out in triplicate. The water kefir grain mass increased from 16.4 ± 0.5 to 28.6 ± 0.6 g during the first 24 h of the fermentation; thereafter, the mass remained constant. The water kefir dry mass initially increased from 13.8 ± 0.1 % (m m\(^{-1}\)) at 0 h (inoculum not yet added to the water kefir simulation medium) to 16.7 ± 0.2 % (m m\(^{-1}\)) after 3 h of fermentation. Hereafter, the dry mass decreased until it remained stable at 13-14 % (m m\(^{-1}\)) (Figure 1 A).

The initial pH of the water kefir simulation medium was 4.85 ± 0.01. This value dropped to 4.26 ± 0.03 after the addition of the water kefir grains at 0 h. After 72 h of fermentation, the pH reached 3.45 ± 0.01, whereafter the pH decreased only slowly to reach 3.35 ± 0.01 after 192 h of fermentation (Figure 1 B).

Microbial enumerations

The viable counts of the water kefir liquor and water kefir grains on the MRS and YG agar media remained constant during the whole fermentation process. Immediately after the water
Kefir grains were added to the water kefir simulation medium and the bottle was mildly turned, the viable counts of the lactic acid bacteria and the yeasts in the water kefir liquor plateaued to a certain level. The average viable counts of yeasts and lactic acid bacteria were 6.3 ± 0.2 and 6.9 ± 0.1 log cfu ml⁻¹ of water kefir liquor, respectively, and 7.4 ± 0.1 and 8.2 ± 0.1 log cfu g⁻¹ of water kefir grains, respectively. A quantifiable number of acetic acid bacteria (>30 colonies on the agar medium with the lowest dilution) could only be found at 144 h and 192 h on the mDMS agar media from the water kefir liquor, namely 3.8 ± 0.1 and 6.2 ± 0.1 log cfu ml⁻¹, respectively. In the case of the water kefir grains, acetic acid bacteria could only be quantified after 192 h of fermentation, namely 4.6 ± 0.1 log cfu g⁻¹. No colonies were found on the KAA and the VRBG agar media, indicating the absence of enterococci plus streptococci and Enterobacteriaceae, respectively.

The ratios of the viable counts of lactic acid bacteria to those of yeasts were also relatively constant during the whole fermentation, with averages of 4.4 ± 1.2 and 6.1 ± 2.4 in the water kefir liquor and on the water kefir grains, respectively, indicating that there were 2 to 10 lactic acid bacteria cells for each yeast cell, both in the water kefir liquor and on the water kefir grains. The ratios of the viable counts of lactic acid bacteria and yeasts on the water kefir grains (cfu g⁻¹) to those in the water kefir liquor (cfu ml⁻¹) were constant too, with averages of 20.4 ± 8.4 and 14.7 ± 4.5, respectively, indicating that the cell density was 10 to 30 times higher on the water kefir grains compared to the water kefir liquor. If the amount of water kefir grains and the amount of water kefir liquor was taken into account, the ratios of the total amounts of cells on the water kefir grains (cfu) to those in the water kefir liquor (cfu) were again relatively constant with averages of 8.8 ± 1.6 and 6.5 ± 1.5 for lactic acid bacteria and yeasts, respectively, indicating that there were 4 to 10 times more micro-organisms on the water kefir grains than in the water kefir liquor. However, because the water kefir grain mass, with higher viable counts compared to the water kefir liquor, increased in mass as a function
of time, there was an overall increase of the total cell count during the first 48 h of the fermentation.

**Culture-dependent species diversity analysis and community dynamics**

The culture-dependent bacterial species diversity analysis revealed the presence of *Lactobacillus casei/paracasei, Lactobacillus hilgardii, Lactobacillus harbinensis, Lactobacillus nagelii, Acetobacter lovaniensis/fabarum, and Lactobacillus hordei/mali* (in decreasing order), of which the first 3 were most dominant (Figure 2). This bacterial species diversity was similar in the water kefir liquor and on the water kefir grains and was more or less constant as a function of the fermentation time (data not shown). All mDMS agar isolates picked up after 192 h of fermentation were identified as *Acetobacter lovaniensis/fabarum*. The culture-dependent yeast species diversity analysis revealed the presence of two species, namely *Saccharomyces cerevisiae* and *Dekkera bruxellensis* (Figure 2). Also the yeast species diversity was similar in the water kefir liquor and on the water kefir grains and was stable during the whole fermentation (data not shown).

**Culture-independent species diversity analysis and community dynamics**

Via the culture-independent assessment of the microbial species diversity, it was confirmed that the three independent fermentations displayed identical PCR-DGGE profiles with the primers used (V3, LAC, Bif, and Yeast) at 0, 24, 72, and 192 h (data not shown). With the V3 primer pair, *Lb. hordei/mali, Lb. harbinensis/brevis, Lb. casei/paracasei/rhamnosus, Lb. hilgardii/diolivorans, Lb. nagelii/ghanensis, B. psychraerophilum/crudilactis*, and a species of Acetobacteraceae were found (Figure 3). The latter appeared in the PCR-DGGE profile (V3 primer pair) of the water kefir liquor after 192 h of fermentation (for the three replicates), but this band was not present in the samples of the water kefir grains (Figure 3).
The presence of the lactic acid bacteria detected with the universal V3 primers was confirmed with the use of the LAC primer pair. With this primer pair, *Lb. hordei/mali, Lb. nagelii/ghanensis, Lb. casei/paracasei, Lb.hilgardii/diolivorans, and Lb. harbinensis/brevis* were found. PCR-DGGE with the Bif primer pair was used to confirm the presence of bifidobacteria in water kefir fermentation. Only one band, identified as *B. psychraerophilum* (96 % identity, Accession No. AB437351), could be found in this gel (data not shown). However, this sequence was very similar to the sequence of an uncultivated *Bifidobacterium* species (100 % identity, Accession No. HE804184) found in water kefir grains. This band was present during the whole fermentation process in the water kefir liquor as well as on the water kefir grains. The PCR-DGGE analysis with the yeast primer pair confirmed the culture-dependent results. Both *S. cerevisiae* (100 % identity, Accession No. JX068683) and *D. bruxellensis* (100 % identity, Accession No. AY969049) were detected during the whole fermentation process. The PCR-DGGE profiles with the LAC, Bif, and yeast primer pairs showed no evolution of the species diversity as a function of time during the whole water kefir fermentation process.

**Substrate consumption and metabolite production**

Sucrose was the main substrate present at the start of the fermentation (0 h). The concentration of sucrose decreased fast from 47.5 ± 1.7 g l⁻¹ at 0 h until 1.2 ± 0.8 g l⁻¹ after 24 h of fermentation. This decrease in sucrose concentration gave rise to an increase in fructose concentration, which reached a maximum after 24 h of fermentation. This was in contrast with the glucose concentration, which decreased continuously during the fermentation. After 72 h, most of the carbohydrates were consumed, with only 3.1 ± 1.0 g l⁻¹ of total carbohydrates left of the initial 75.1 ± 2.1 g l⁻¹ (Figure 1 B).
The ethanol concentration increased linearly from $1.1 \pm 0.1 \text{ g l}^{-1}$ at 0 h to $20.3 \pm 1.3 \text{ g l}^{-1}$ after 72 h. In this time frame, the lactic acid concentration increased from $0.7 \pm 0.1 \text{ g l}^{-1}$ to $4.9 \pm 0.2 \text{ g l}^{-1}$ and the acetic acid concentration increased from $0.1 \pm 0.0 \text{ g l}^{-1}$ to $1.0 \pm 0.1 \text{ g l}^{-1}$. The glycerol and mannitol concentrations reached $2.3 \pm 0.2 \text{ g l}^{-1}$ and $0.8 \pm 0.0 \text{ g l}^{-1}$ after 72 h of fermentation, respectively (Figure 1 C).

Only a limited amount of volatile aroma compounds (besides acetic acid and ethanol) could be found with the SH-GC-MS method used. These compounds were ethyl acetate, 2-methylpropanol, isoamyl alcohol, isoamyl acetate, ethyl hexanoate, ethyl octanoate, and ethyl decanoate. Also ethyl butanoate and ethyl 2-methyl-butanoate were found, but these two compounds could not be quantified because their concentrations were too close to the quantification limit of the method used. Except for ethyl acetate and ethyl decanoate, the production of these compounds stopped as soon as the carbohydrates were exhausted (after 72 h of fermentation). The production of ethyl acetate and ethyl decanoate, however, continued until 192 h and 96 h of fermentation, respectively (Figure 4). Compared to their threshold levels, the most dominant aroma components of this water kefir after 72 h were the esters (Table 1). In particular, ethyl octanoate was present at 688 times of its threshold value.

### Carbon recovery

After 192 h of fermentation, a carbon recovery of 100.6 % was obtained, indicating that all major substrates and metabolites were recovered from this water kefir fermentation. After 72 h of fermentation, the majority of the metabolic activity had taken place. The major end products of the fermentation were ethanol, carbon dioxide, lactic acid, glycerol, mannitol, and acetic acid, besides the synthesis of water kefir grain mass (Figure 5).

### Discussion
The multiphasic microbial approach of the present study revealed lactic acid bacteria, yeasts, bifidobacteria, and acetic acid bacteria as the main micro-organisms present during water kefir fermentation (2, 4, 19, 20). The species diversity was similar in the water kefir liquor and on the water kefir grains and remained stable in both phases during the whole fermentation process. However, the density of the micro-organisms was higher on the water kefir grains than in the water kefir liquor, indicating that the water kefir grains harbored the micro-organisms. This finding also explains the absence of an increasing microbial population during the water kefir fermentation process. Although the yeast metabolism predominated the fermentation, lactic acid bacteria were present in higher numbers than yeasts. As common fermentation times for water kefir are around 72 h, the acetic acid bacteria, appearing only after 144 h of fermentation, were not an important part of the water kefir ecosystem of the present study. Communities of acetic acid bacteria found in water kefir usually range from negligible (10) to > 8 log cfu ml\(^{-1}\) (4). The absence of enterococci and Enterobacteriaceae was to be expected, considering the fast decrease of the pH during water kefir fermentation to a pH below 3.5.

The most dominant lactic acid bacteria (in decreasing order) were *Lb. casei*/*paracasei*, *Lb. hilgardii*, and *Lb. harbinensis*. The former two species have been associated with water kefir before (1, 2, 4, 5, 11, 13, 14, 18, 19, 38). To our knowledge, this is the first time that *Lb. harbinensis* was found in water kefir. This facultative heterofermentative lactic acid bacterium was first isolated from a Chinese vegetable fermentation (39), and has been found afterwards in French cow milk (40), the oral ecosystem of healthy individuals (41), Parmigiano Reggiano cheese (42), and sorghum sourdough fermentation (43). It is worth to notice that *Lb. harbinensis* produces antifungal compounds that inhibit yeasts (40, 44). *Lactobacillus casei*, also a facultative heterofermentative lactic acid bacterium, is also associated with the oral ecosystem of healthy individuals, the human intestinal tract, and raw
and fermented dairy and vegetable products (45). Some strains of Lb. casei show probiotic potential (46), which makes the water kefir ecosystem a possible source of novel probiotic Lb. casei strains. Lactobacillus hilgardii, an obligate heterofermentative lactic acid bacterium, occurs also in wine and cocoa fermentations (47, 48). This bacterium is believed to be the main exopolysaccharide (EPS) producer in the water kefir ecosystem (5, 18, 19, 38). However, not all Lb. hilgardii strains from water kefir produce EPS (4). Other EPS-producing lactic acid bacteria isolated from water kefir include Leuconostoc mesenteroides, Lb. brevis, Lb. casei, Lb. nagelli, and Lb. hordei (4, 7). The water kefir of the present study harbored Lb. nagelli, a homofermentative lactic acid bacterium frequently found in water kefir (1, 2, 4). For homopolysaccharide production, sucrose is the necessary substrate (49). In the water kefir of the present study, the water kefir grain mass was only produced in the early stages of the fermentation, as long as sucrose was present. The accumulation of fructose in the water kefir liquor indicates that the water kefir grains were composed of glucan. The homopolysaccharide produced did not serve as a reserve polymer, as the kefir grain (dry) mass did not decrease upon prolonged fermentation.

The detection of B. psychraerophilum/crudilactis confirms a recent finding that water kefir harbors bifidobacteria (2, 3, 20). This species was first isolated from a porcine caecum (50). Bifidobacteria are obligate anaerobic bacteria that produce more acetate than lactate. Because of the low acetate concentrations of the water kefir of the present study, their metabolic activity was limited.

The most dominant yeast species was S. cerevisiae. This yeast species is frequently associated with water kefir (3, 4, 10, 13, 14, 19, 20, 51). Dekkera bruxellensis (anamorph Brettanomyces bruxellensis) was only recently shown to be associated with water kefir (3, 20). This yeast plays a key role in the spontaneous fermentation of typical Belgian acid ales (52), although it is usually associated with spoilage of beer and wine (53). Whether the presence of D.
*bruxellensis* during water kefir fermentation has a positive or negative influence on the end product is unclear at this moment.

The wide meta-metabolomics approach of the present study elucidated the substrate consumption and metabolite production profiles of the microbial consortium described above. The major metabolites were ethanol, carbon dioxide, and lactic acid. Lactic acid, responsible for the fresh sour taste of water kefir, was the main metabolite of the lactic acid bacteria species, although smaller amounts of ethanol, acetate, and mannitol were produced too. Although there were high concentrations of fructose present at the initial stage of the fermentation process, the production of mannitol from fructose by the heterofermentative lactic acid bacteria species was limited, but could explain part of the acetate production. Mannitol has a sweet taste and possesses antioxidant activity (54); both properties might be desirable in water kefir. Ethanol and carbon dioxide were the main metabolites produced by the yeasts, although smaller amounts of glycerol and acetic acid were produced too. Glycerol is a slightly sweet molecule that may slightly increase the viscosity of a fermented beverage, but that does not seem to have a direct influence on the taste and aroma of fermented beverages (55).

All esters and higher alcohols found in the water kefir liquor are associated with yeast metabolism (56). For instance, hexanoic acid, octanoic acid, and decanoic acid, necessary for the production of the corresponding ethyl esters, originate from the fatty acid biosynthesis pathway in yeasts. All these volatile aroma compounds are also found in wine and beer, but a direct comparison of the water kefir liquor with beer or wine is difficult, because of the multitude of interactions between all the chemical components in each fermented beverage. This makes it also difficult to estimate the impact of individual aroma compounds on the overall flavor. However, with regards to the threshold values of the different aroma compounds, the esters isoamyl acetate, ethyl hexanoate, ethyl octanoate, and ethyl decanoate
will exert the greatest influence on the aroma of the water kefir of the present study, contributing to its fruity and floral notes.

In conclusion, a sound water kefir fermentation with good growing water kefir grains was obtained during the present study, which can be used as reference for other water kefirs. This water kefir fermentation was dominated by the lactic acid bacteria species *Lb. casei*/*paracasei*, *Lb. harbinensis*, and *Lb. hilgardii*, and the yeasts *S. cerevisiae* and *D. bruxellensis*. More lactic acid bacteria were present compared to yeasts, although the metabolism of the yeasts prevailed. The majority of the micro-organisms was present on the water kefir grains. The water kefir grain mass increased as long as sucrose was present, and the main metabolites produced during the fermentation were ethanol, carbon dioxide, lactic acid, glycerol, and acetic acid. Isoamyl acetate, ethyl hexanoate, ethyl octanoate, and ethyl decanoate dominated the aroma of the water kefir of the present study.
Acknowledgements

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References


Legends to the figures

Figure 1: A: The production of water kefir grain mass (g, ●); and water kefir grain dry mass (%○); as a function of time (h). B: The pH evolution and the consumption of carbohydrates (g l⁻¹) as a function of time (h): pH, ○; total carbohydrates, ●; sucrose, ▲; fructose, ■; and glucose, ●. C: The production of metabolites (g l⁻¹) as a function of time (h): ethanol, ■; lactate, Δ; glycerol, ◊; acetate, ○; and mannitol, □.

Figure 2: The microbial species diversity of the water kefir liquor and water kefir grains, represented by pooling all samplings. The closest relatives to the sequenced fragments are given. Left (bacterial species diversity): 1, Lactobacillus casei/paracasei (99 % identity, Accession No. KF500575/KF516078); 2, Lactobacillus hilgardii (99 % identity, Accession No. JX099894); 3, Lactobacillus nagelii (99 % identity, Accession No. AB370876); 4, Lactobacillus harbinensis (100 % identity, Accession No. KF418816); 5, Acetobacter lovaniensis/fabarum (99 % identity, Accession No. FJ157228/AB665084); 6, Lactobacillus mali (99 % identity, Accession No. AB326352); and 7, unknown. Right (yeast species diversity): 1, Saccharomyces cerevisiae [LSU (99 % identity, Accession No. JQ914745) and ITS (99% identity, Accession No. KC515374)]; and 2, Dekkera bruxellensis [LSU (100% identity, Accession No.: JQ689028) and ITS (100% identity, Accession No. FJ545249)].

Figure 3: The PCR-DGGE-profiles of the water kefir liquor (L) and the water kefir grains (G) after 0, 24, 72, and 192 h (indicated as subscripts) of fermentation. The closest relatives to the fragments sequenced are given. A: Use of the universal V3 primer pair: 1, 2, 3, 5, 7: Lactobacillus hordei/mali (98 % identity, Accession No. NR044394/AB326352); 4: unspecified bacterium; 6: Lactobacillus harbinensis/brevis (100 % identity, Accession No. KF418816/AY974809); 8, 9, 13, 14, 19, 20: Lactobacillus casei/rhamnosus/paracasei (100 % identity, Accession No. KF500575/KF554252/KF516078); 10, 16: Lactobacillus
hilgardii/diolivorans (100 % identity, Accession No. KF418826/KF149643); 11, 17: 641 
Lactobacillus ghanensis/nagelii (97 % identity, Accession No. AB690235/AB370876); 12, 642 
18: Bifidobacterium psychraerophilum (98 % identity, Accession No. AB437351); and 15: 643 
Acetobacteraceae. B: Use of the LAC primer pair: 5, 7, 11, 12, 14, 18, 19: Lb. casei/paracasei 644 
(99 % identity, Accession No. KF500575/KF516078); 4, 8, 9, 15, 16: Lb. nagelii (99 % 645 
identity, Accession No. AB370876); 1, 2, 3: Lb. hordei/mali (99 % identity, Accession No. 646 
NR044394/AB326352); 6, 13: Lb. harbinensis/brevis (99 % identity, Accession No. 647 
KF418816/AY974809); and 10, 17: Lb. hilgardii/diolivorans (99 % identity, Accession No. 648 
KF418826/KF149643).

Figure 4: A: The production of volatile short-chain aroma compounds (mg l⁻¹) as a function 650 
of time (h): isoamyl acetate, ●; isoamyl alcohol, ○; ethyl acetate, ◊; and 2-methyl-1-propanol, □. 651  
B: The production of volatile long-chain aroma compounds (mg l⁻¹) as a function of time 652 
(h): ethyl octanoate, □; ethyl decanoate, Δ; and ethyl hexanoate, ◊.

Figure 5: Presence of different carbon-containing constituents of the water kefir fermentation 654 
process, as a function of time (h), expressed as percentage (%) of the total amount of carbon 655 
recovered: 1, substrates; 2, kefir grain mass; 3, ethanol; 4, carbon dioxide; 5, glycerol; 6, 656 
mannitol; 7, lactate; and 8, acetate.

658
Table 1: Concentrations of volatile aroma compounds found after 72 h of water kefir fermentation. The Kovats index (KI) and the method of identification (Id) are given for every compound. Identification was via the mass spectrum (MS) and by comparison with the retention time of the reference compound (RF).

<table>
<thead>
<tr>
<th>Volatile aroma compound</th>
<th>Concentration after 72 h (mg l⁻¹)</th>
<th>KI</th>
<th>Id</th>
<th>Threshold valueb (mg l⁻¹)</th>
<th>Aroma descriptorc</th>
</tr>
</thead>
<tbody>
<tr>
<td>2-Methyl-1-propanol</td>
<td>11.62 ± 0.05</td>
<td>1097</td>
<td>MS/RF</td>
<td>40</td>
<td>Spirituous, fuel</td>
</tr>
<tr>
<td>Isoamyl alcohol</td>
<td>44.13 ± 0.82</td>
<td>1222</td>
<td>MS/RF</td>
<td>30</td>
<td>Harsh, nail polish remover</td>
</tr>
<tr>
<td>Ethyl acetate</td>
<td>13.40 ± 0.58</td>
<td>831</td>
<td>MS/RF</td>
<td>7.5</td>
<td>Fruity</td>
</tr>
<tr>
<td>Isoamyl acetate</td>
<td>0.11 ± 0.01</td>
<td>1141</td>
<td>MS/RF</td>
<td>&gt; 150</td>
<td>Varnish, nail polish remover</td>
</tr>
<tr>
<td>Ethyl hexanoate</td>
<td>0.37 ± 0.01</td>
<td>1250</td>
<td>MS/RF</td>
<td>0.03</td>
<td>Sweet, fruity, banana, pear</td>
</tr>
<tr>
<td>Ethyl octanoate</td>
<td>3.44 ± 0.61</td>
<td>1450</td>
<td>MS/RF</td>
<td>0.005</td>
<td>Fruity, pineapple, pear</td>
</tr>
<tr>
<td>Ethyl decanoate</td>
<td>1.40 ± 0.18</td>
<td>1659</td>
<td>MS/RF</td>
<td>0.2</td>
<td>Floral</td>
</tr>
<tr>
<td>Ethyl butanoate</td>
<td>DNQa</td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ethyl 2-methyl-butanoate</td>
<td>DNQa</td>
<td>1043</td>
<td>MS</td>
<td>0.02</td>
<td>Floral, fruity</td>
</tr>
</tbody>
</table>

a Detected but not quantified.

b For references, see (57, 58).

c For references, see (56, 59-61).